



Advanced Electrocatalysis for Green Hydrogen Production

Ziba S.H.S. Rajan, Genevieve C. Moss, Bamato J. Itota, Darija Susac, and Rhiyaad Mohamed

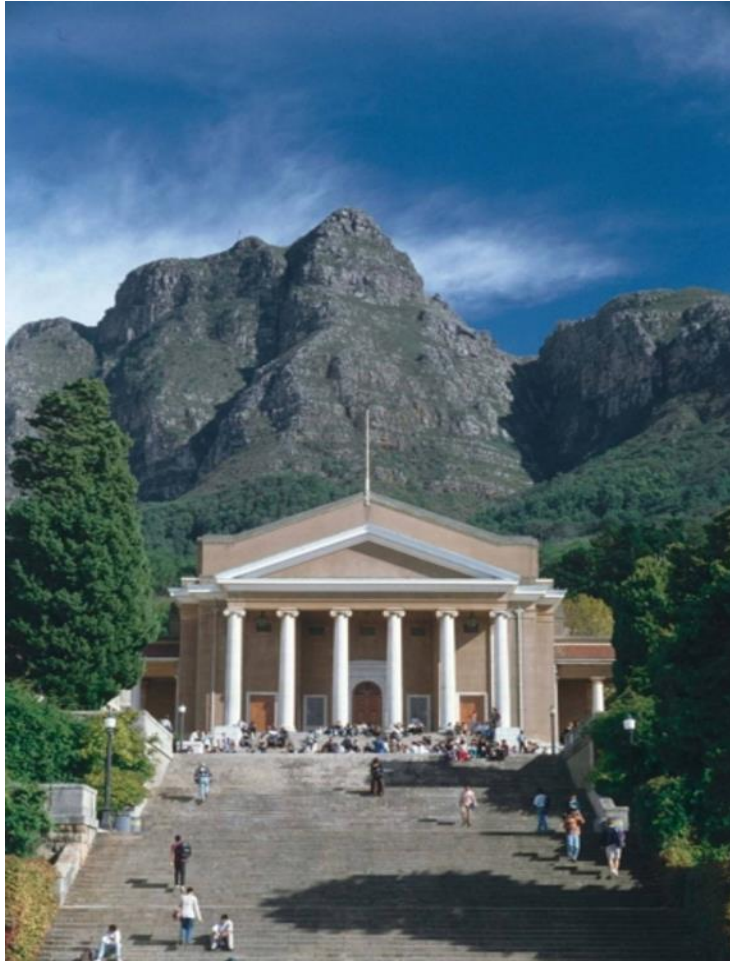
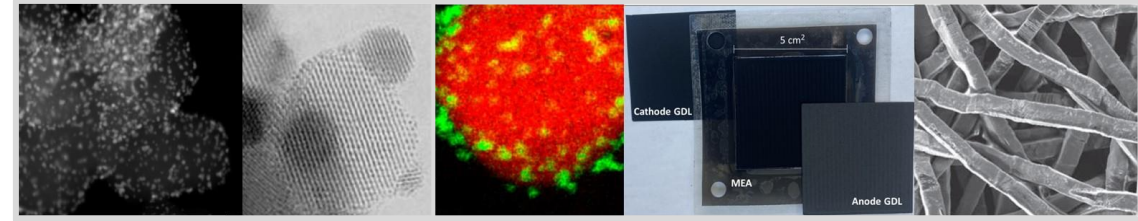
April 7th, 2025

HySA Catalysis

University of Cape Town (UCT)

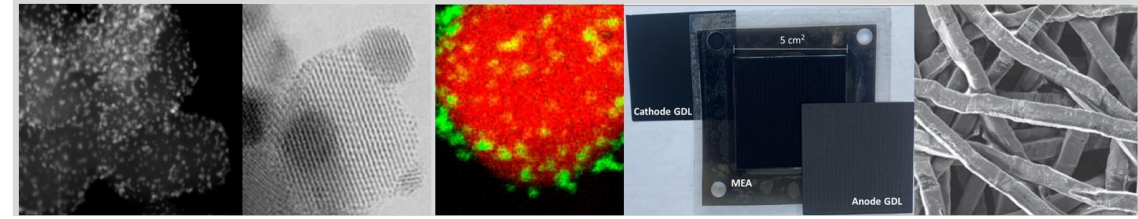
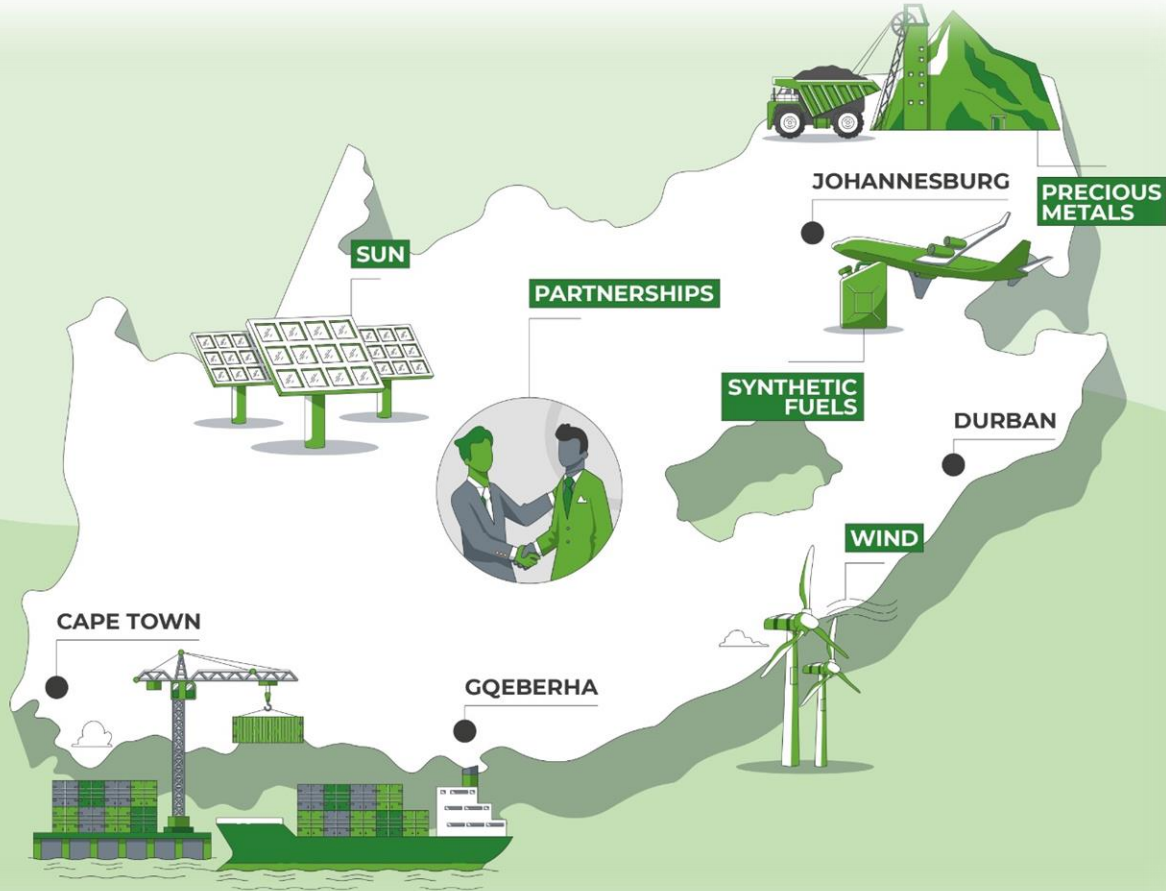
South Africa

<https://ebe.uct.ac.za/catalysis-institute>



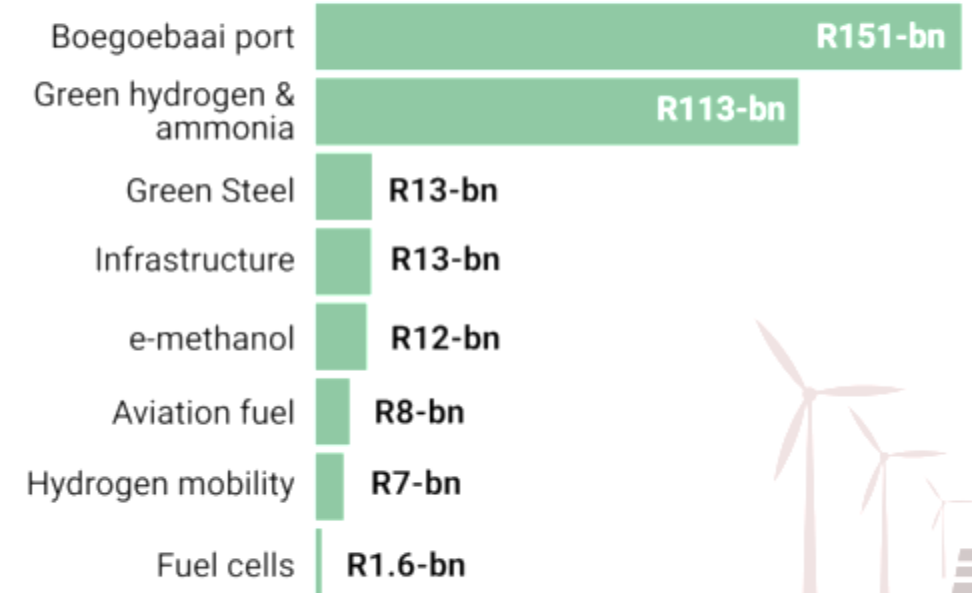
Centre of Competence
Co-hosted by UCT and
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The Big Picture – SA's Potential for Hydrogen



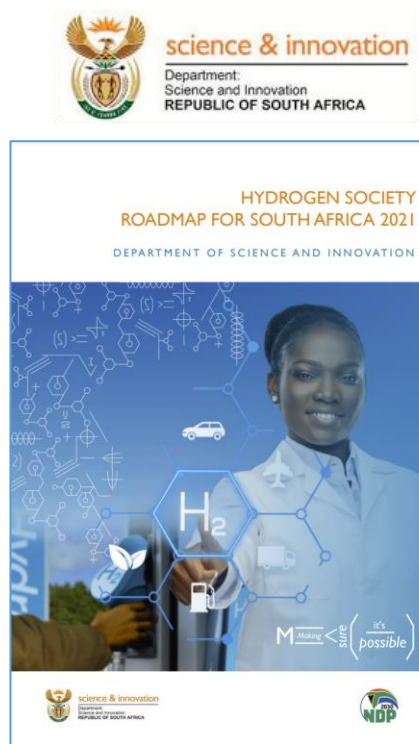
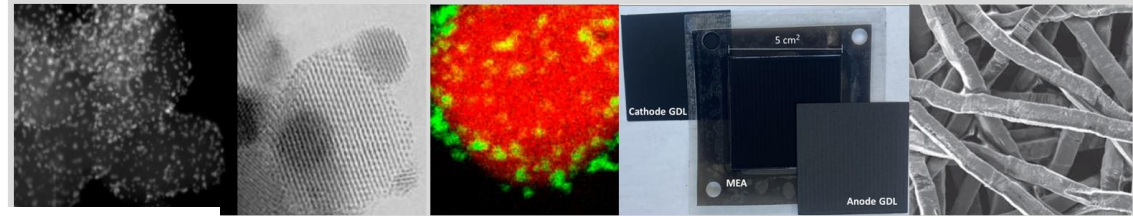
SA's green hydrogen plan

An estimated R319bn is needed to develop the sector, which includes a port and 7 hydrogen projects



Source: Just Energy Transition Investment Plan 2023-2027

The Hydrogen Society Road Map



https://www.dst.gov.za/images/South_Africa_Hydrogen_Society_RoadmapV1.pdf



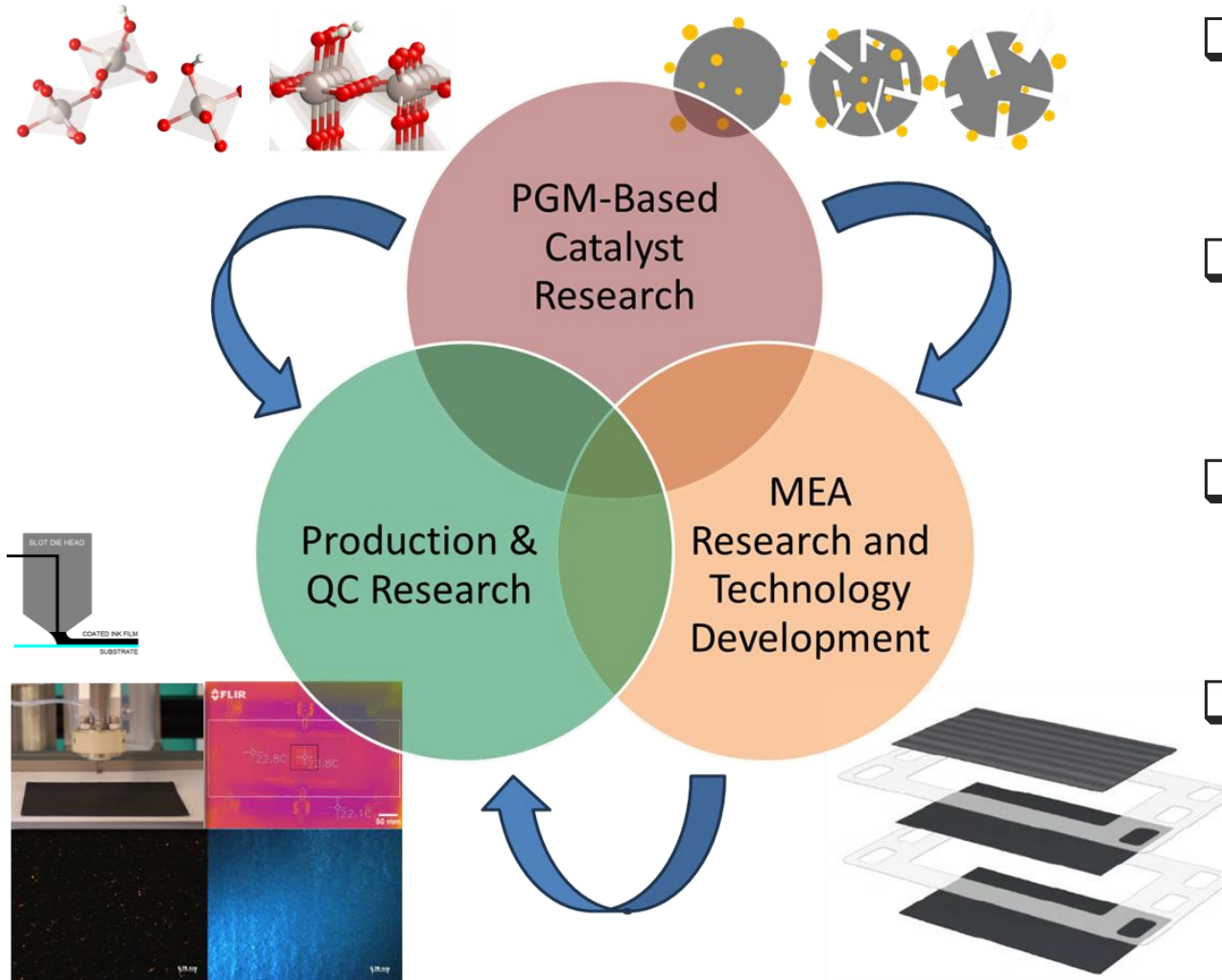
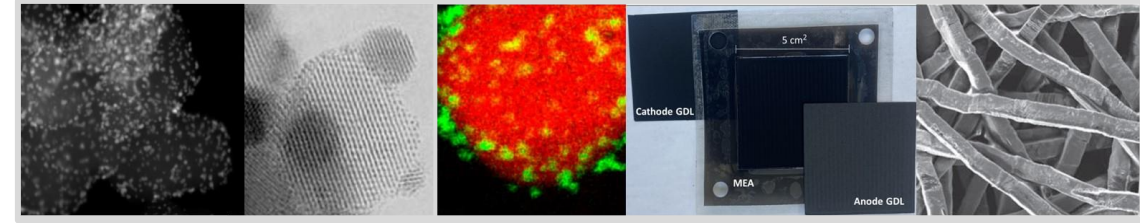
HySA Catalysis Mandate (R&TD):

Components: catalysts and MEAs for PEM fuel cells and electrolyzers

Electrolyzers: H₂ generation

Fuel cells: H₂ utilisation for electricity production

Pem Fuel Cell And Electrolyser Materials: From Catalysts to MEAs (TRL 1 to 6)



- ☐ Optimization of formulations & process technologies
- ☐ Advanced characterisation techniques – *ex-situ* & *in-situ* testing
- ☐ Accelerated stress testing (AST) & durability testing
- ☐ Design lever & transfer function development through structure-property-performance (SPP) correlations

The Research Focus: Electrocatalysis for Energy Applications

- ❑ Water electrolysis to make **green hydrogen**
- ❑ Carbon dioxide reduction to make **valuable chemicals**
- ❑ Hydrogen fuel cells to **generate electricity**

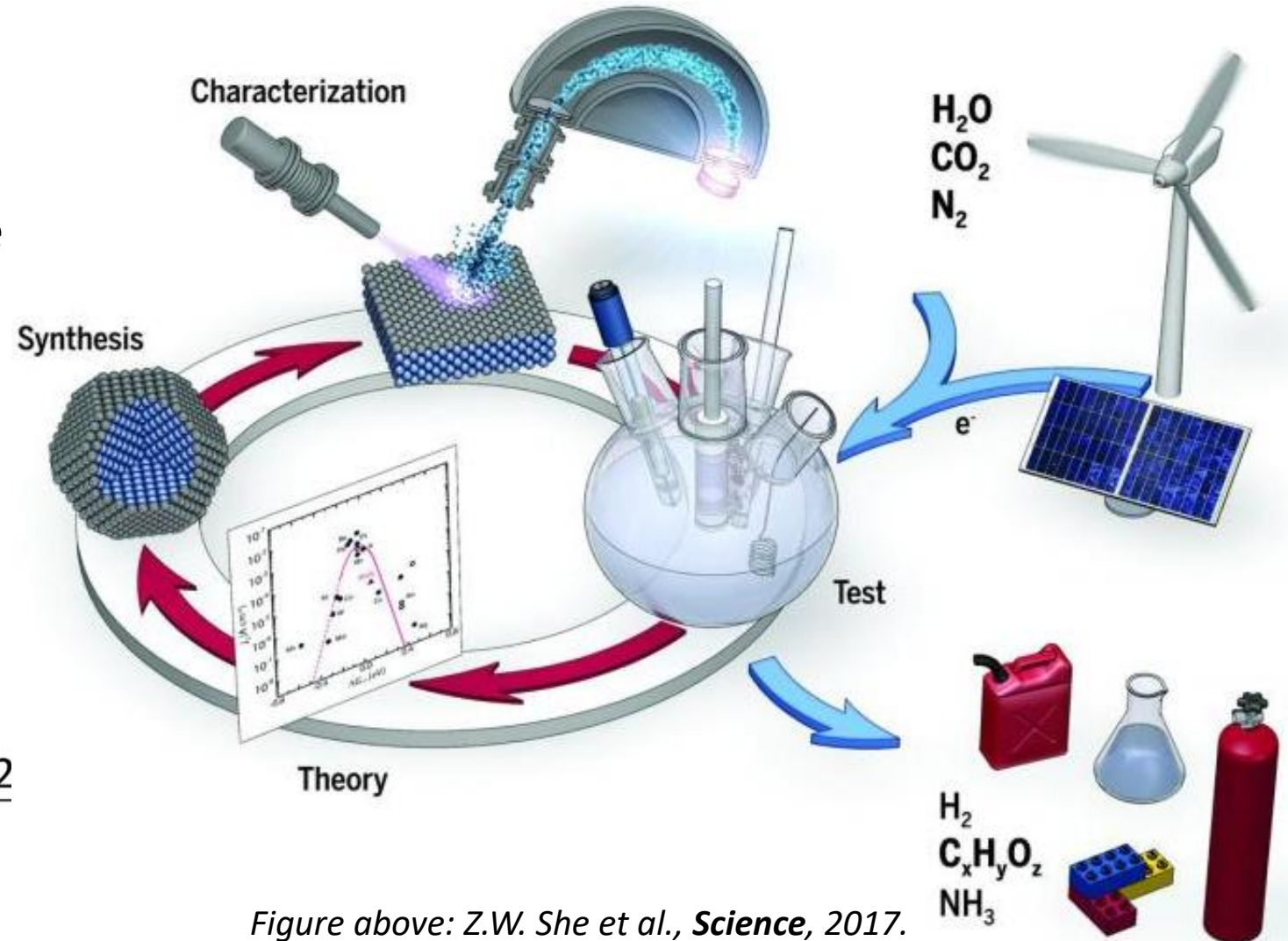
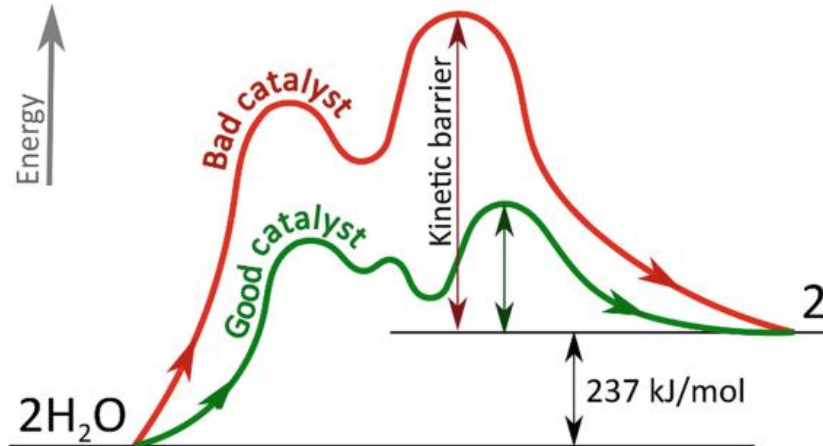
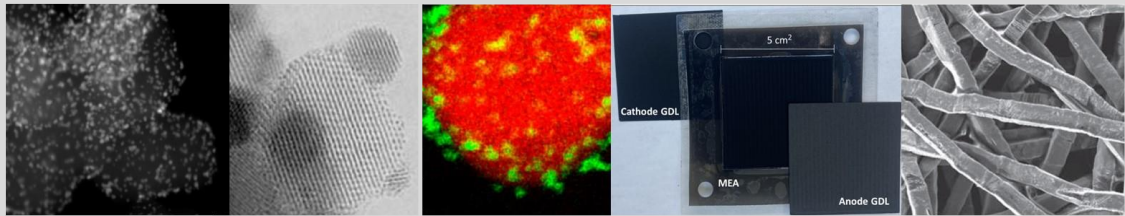
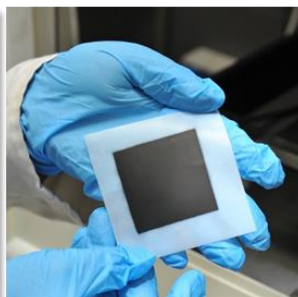
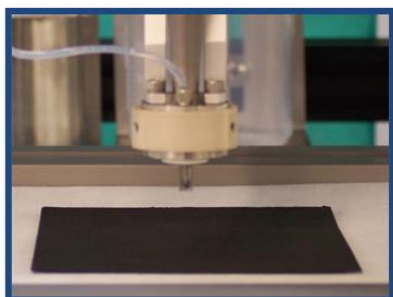


Figure above: Z.W. She et al., *Science*, 2017.

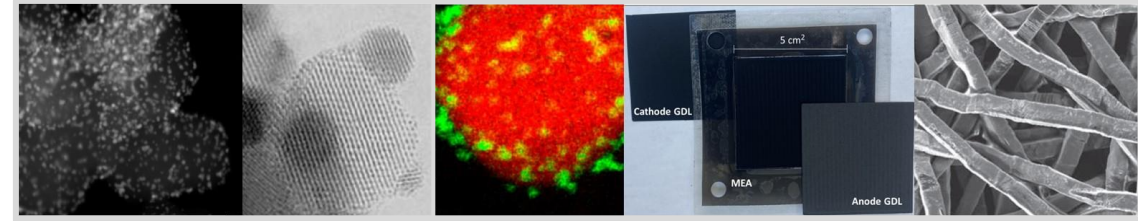
Pem Fuel Cell and Electrolyser Materials: From Catalyst Materials to MEAs (TRLs 1 to 6)



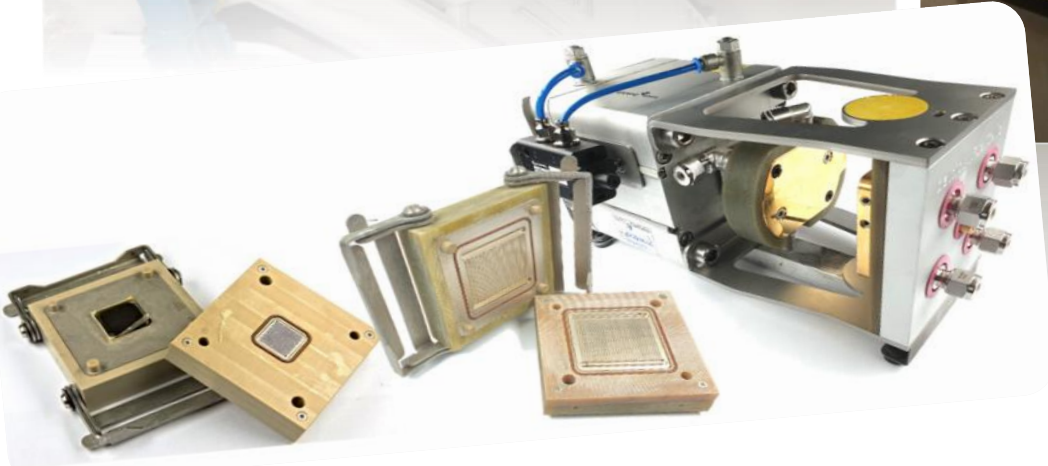
To impact the **global PEM fuel cell and electrolyser market** with **locally developed** catalyst and membrane electrode assembly technologies, advancing the **beneficiation of South Africa's PGMs**



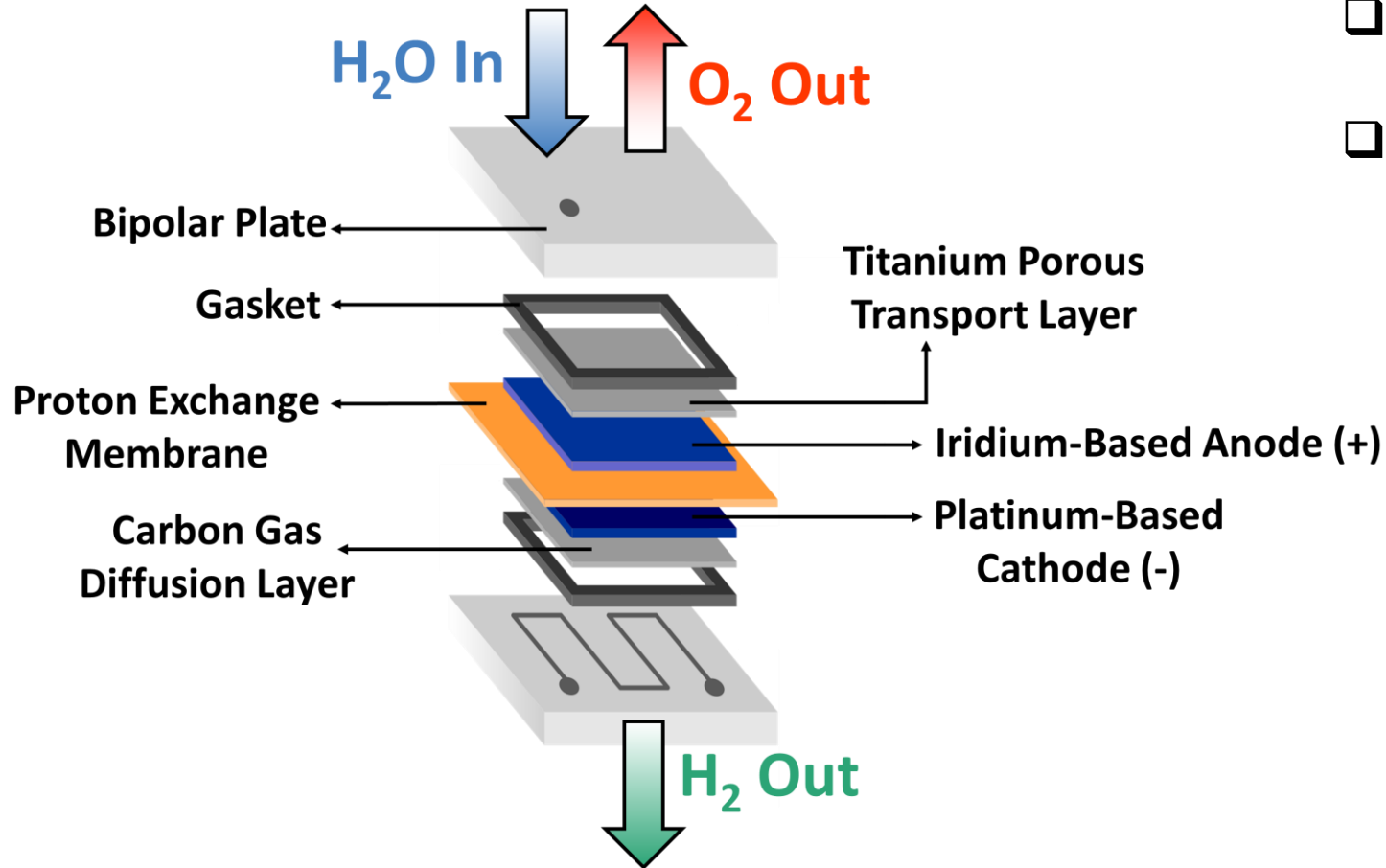
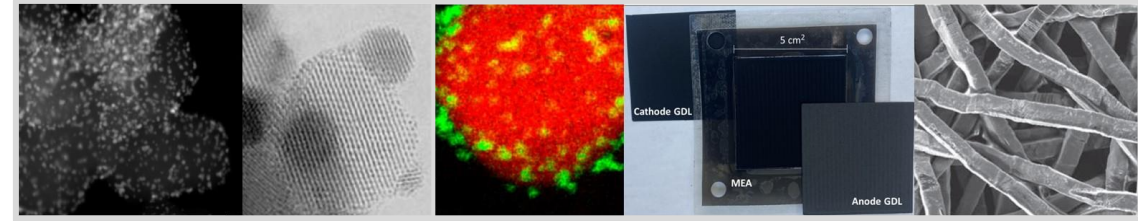
Current Capabilities



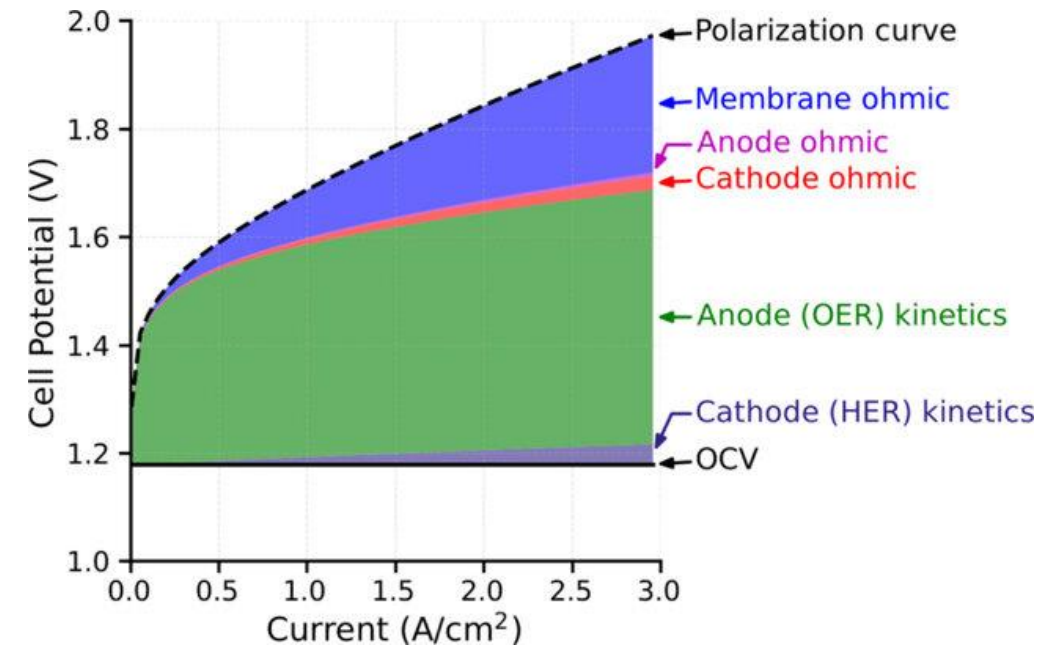
- ❑ Fully equipped for **material synthesis**, characterisation, and screening up to the **25-gram batch size**
- ❑ Electrode **fabrication**, characterisation, and **testing** capabilities



PEM Water Electrolysis

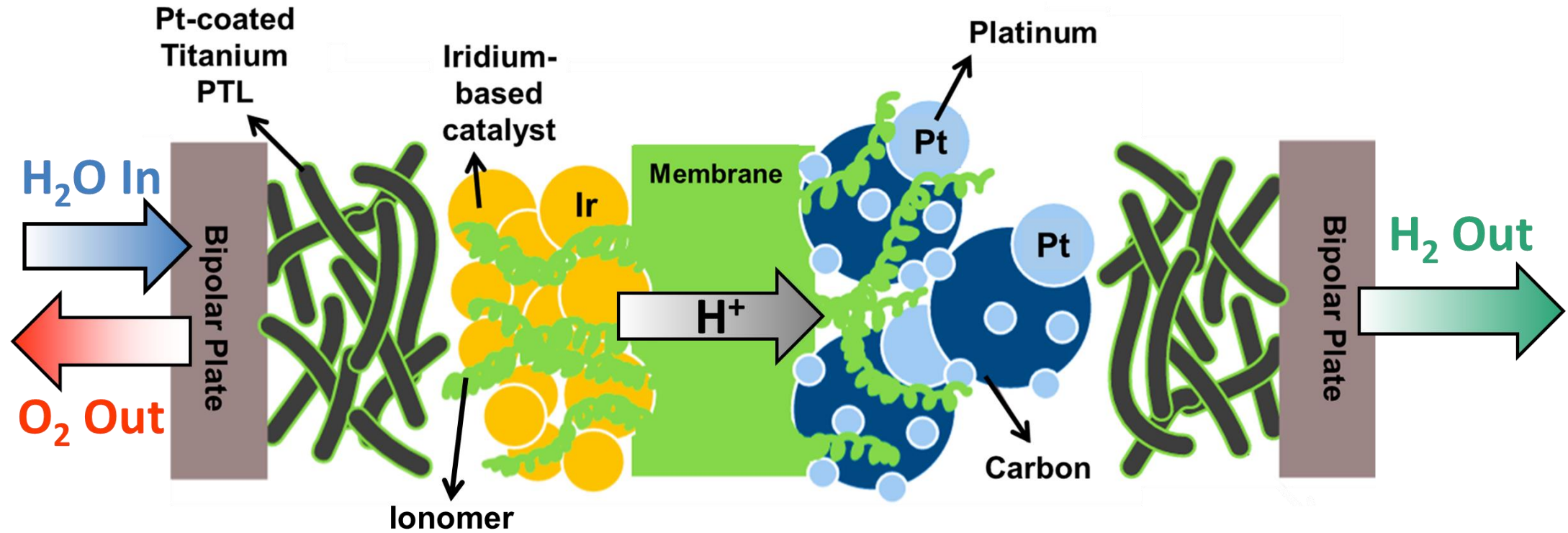
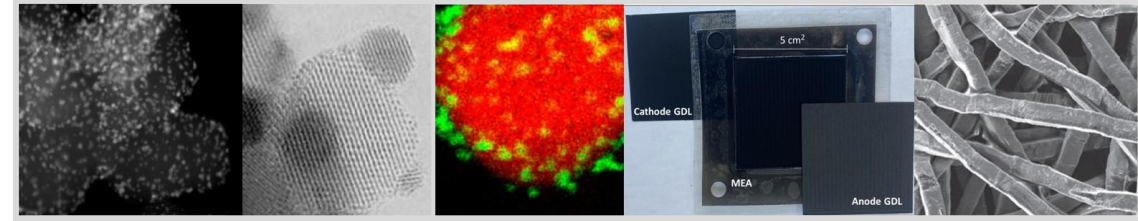


- ❑ Iridium is required to catalyse the **sluggish anodic oxygen evolution reaction (OER)**
- ❑ Platinum is required to catalyse **the cathodic hydrogen evolution reaction (HER)**



Loss Breakdown Figure: M.R. Gerhardt et al., *Journal of the Electrochemical Society*, 2021.

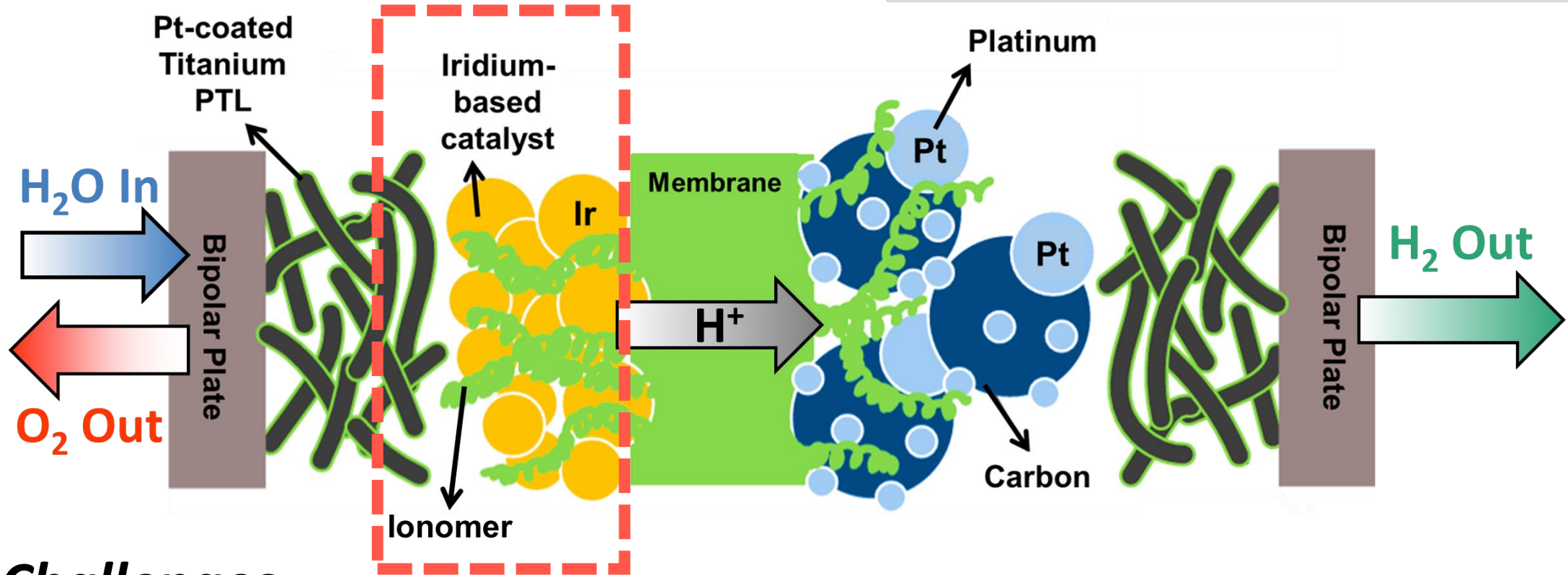
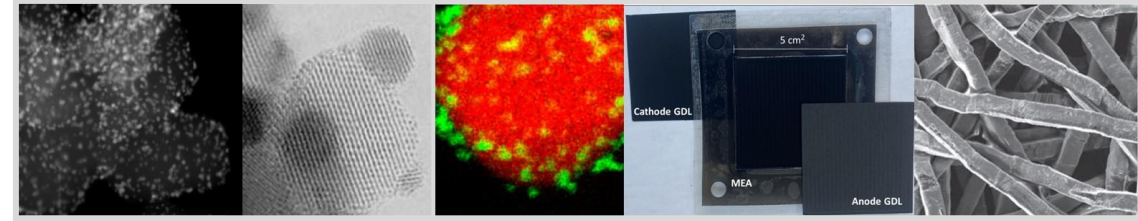
The Membrane Electrode Assembly



Key Challenges:

- ☐ Electrical conductivity of the components (CL/PTL)
- ☐ H^+ transport resistance
- ☐ Hydrogen gas crossover
- ☐ Electrochemical material degradation
- ☐ Anodic catalyst layer PGM content
- ☐ Anodic catalyst layer porosity

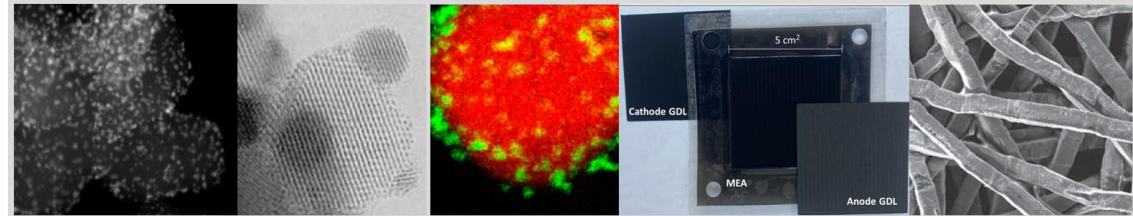
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Key Challenges:

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Closing the Iridium Gap



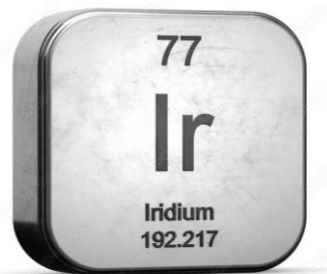
Next-Generation Catalyst Designs:

- ☐ Alloying
- ☐ High-surface-area rutile-type iridium oxides
- ☐ Core-shell concepts
- ☐ Supported Ir-based materials

7 tons with
reduced Ir
loading through
next-generation
catalyst designs

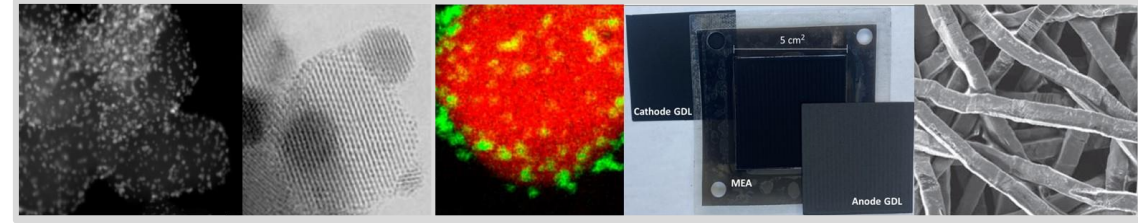
28 tons
required with
traditional OER
catalysts

12 tons available
on the market, with
75% located in South
Africa



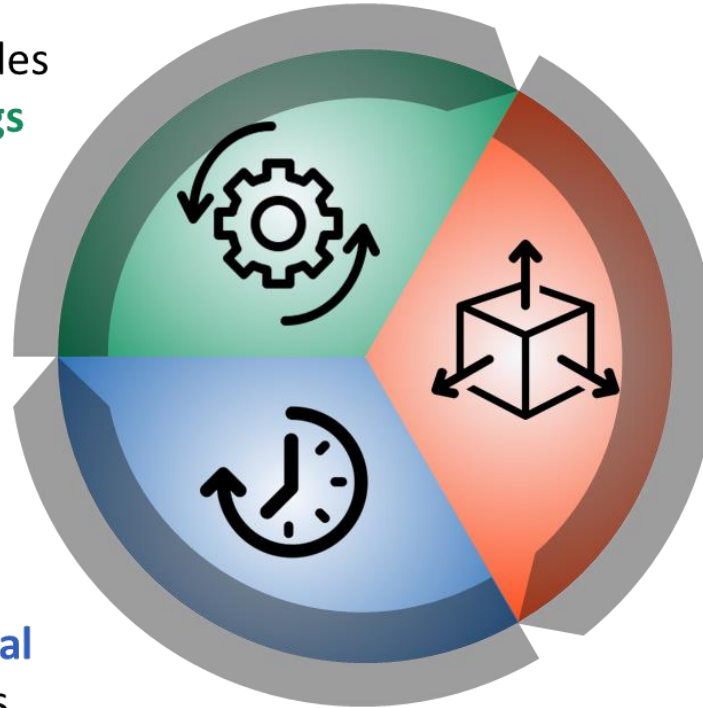
Minke, C. et al., *International Journal of Hydrogen Energy*, 2021.

Key Research & Development Themes



Material Efficiency

- ❑ Enhancing catalytic efficiency of **OER**/HER reactions
- ❑ High-performance electrodes with **reduced PGM loadings**



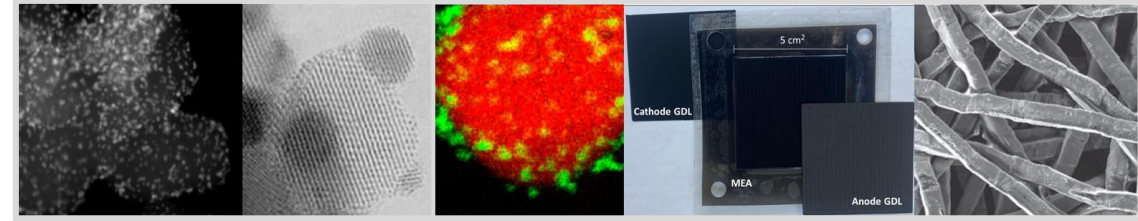
Durability

- ❑ Improving the **electrochemical stability** of OER/HER catalysts
- ❑ Optimised electrode **microstructure**

Scalability

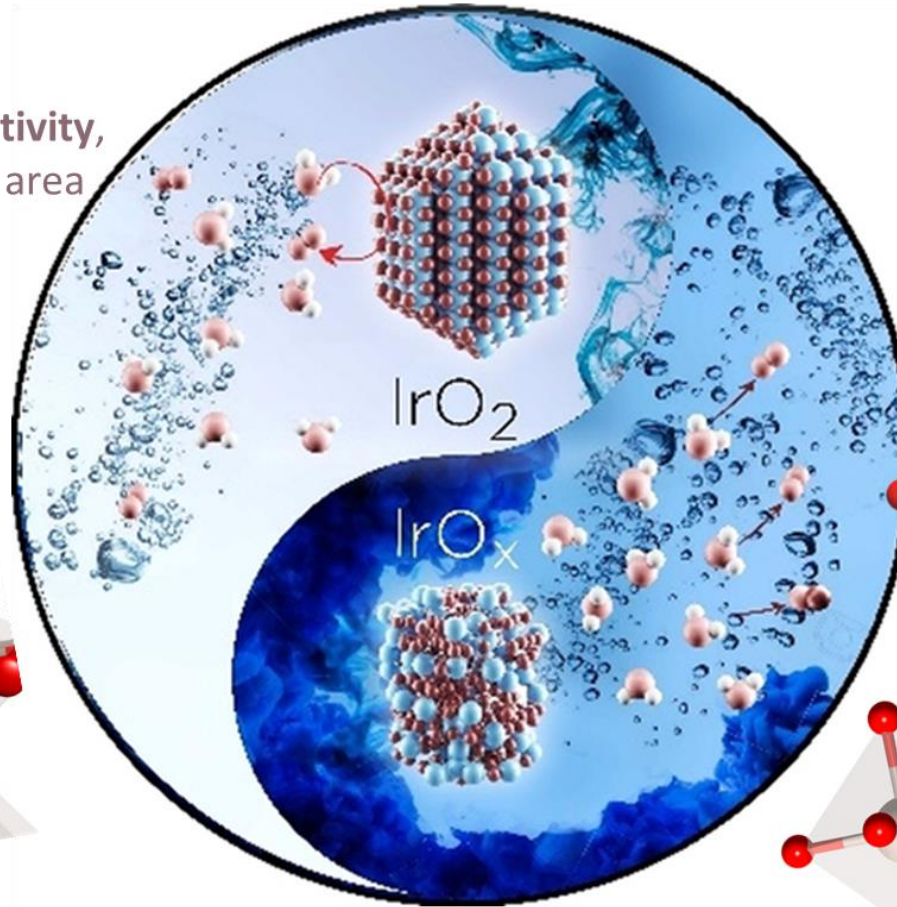
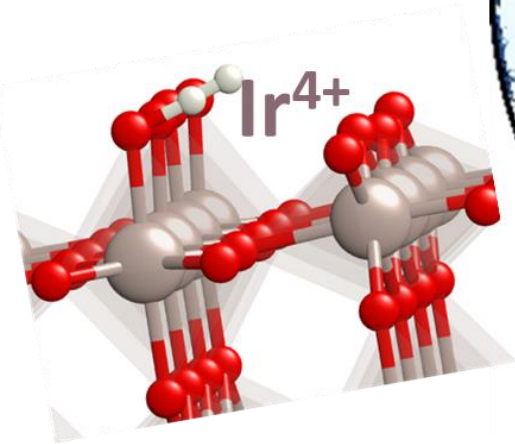
- ❑ Cost-effective, scalable **catalyst designs and preparation approaches**
- ❑ Scalable electrode **fabrication processes**

The OER & the Ir-Oxide Phase Conundrum



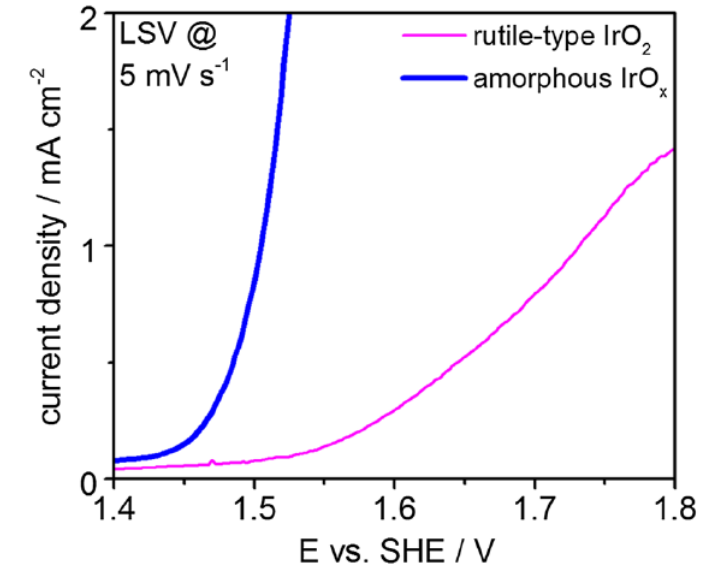
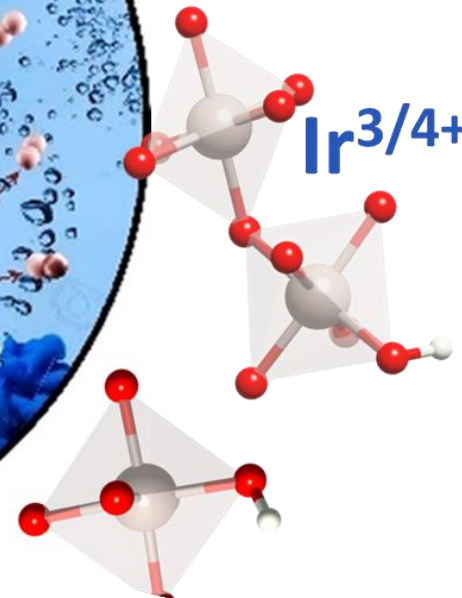
Rutile-type IrO_2

- High stability, less dissolution-prone
- Low intrinsic OER activity, typically low surface area
- High electrical conductivity



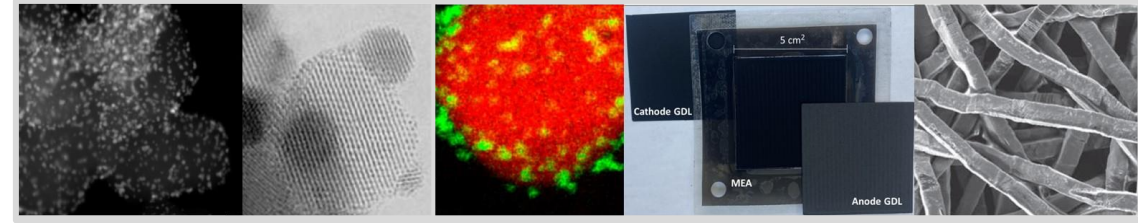
Amorphous IrO_x

- Low stability, highly dissolution-prone
- High intrinsic OER activity, high surface area
- Poor electrical conductivity



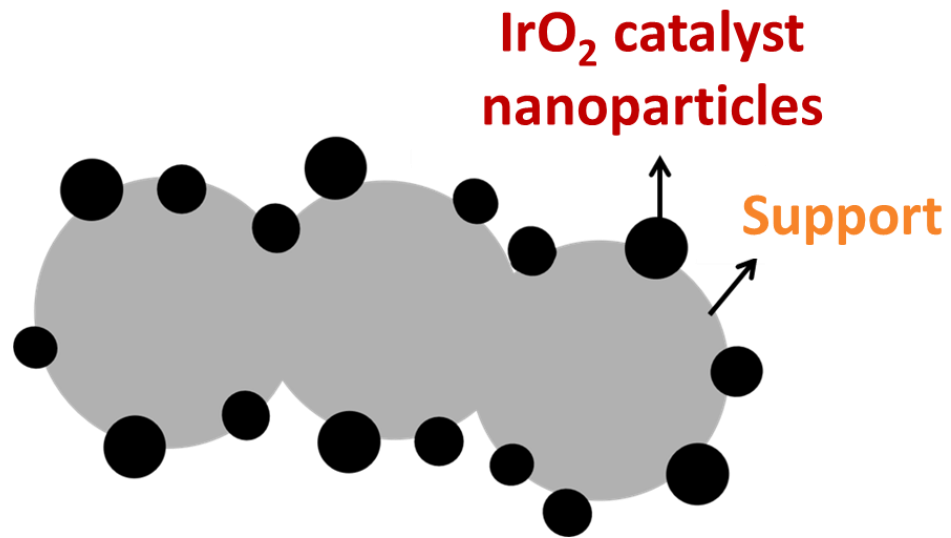
V. Pfeifer et al., *Surface Science*, 2015.

Our Catalyst Solutions

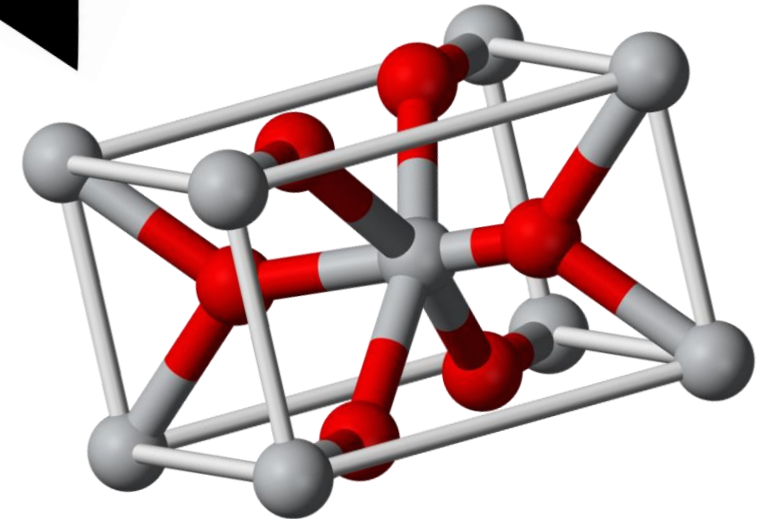


Rutile-type IrO_2 Catalyst Designs

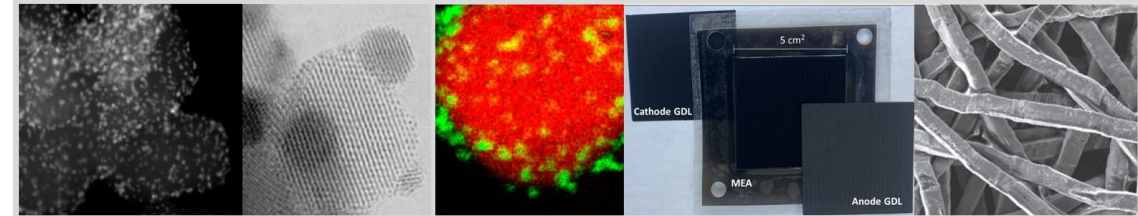
Oxide-Supported Catalysts

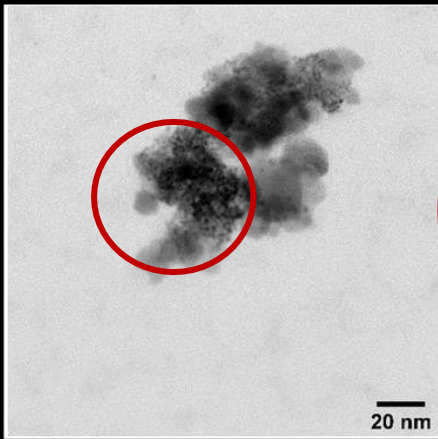
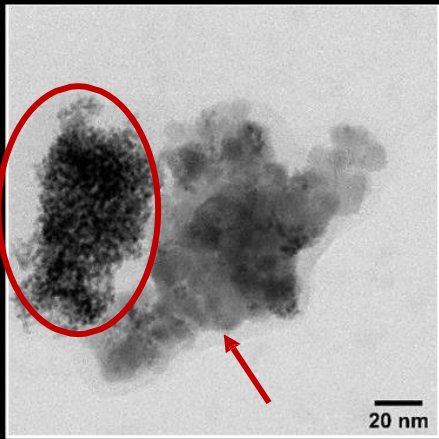
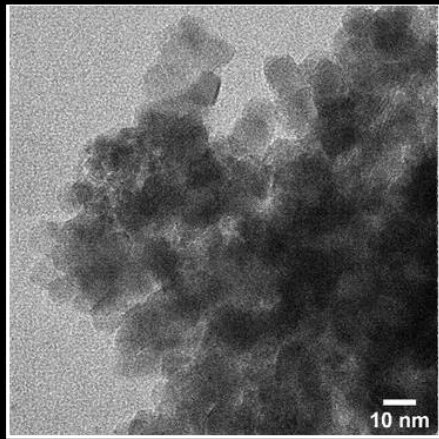


Nanocrystalline IrO_2



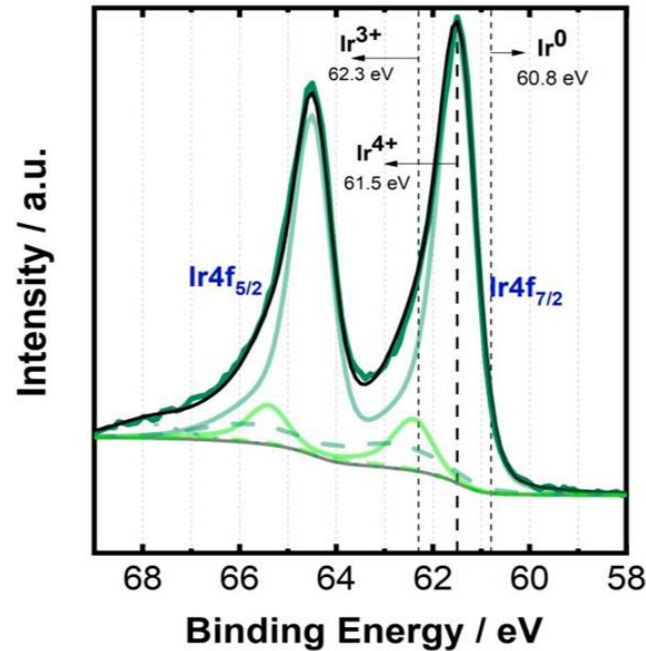
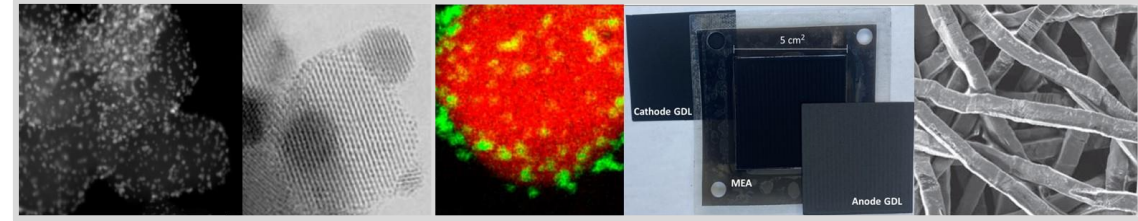
IrO₂ Supported on ATO



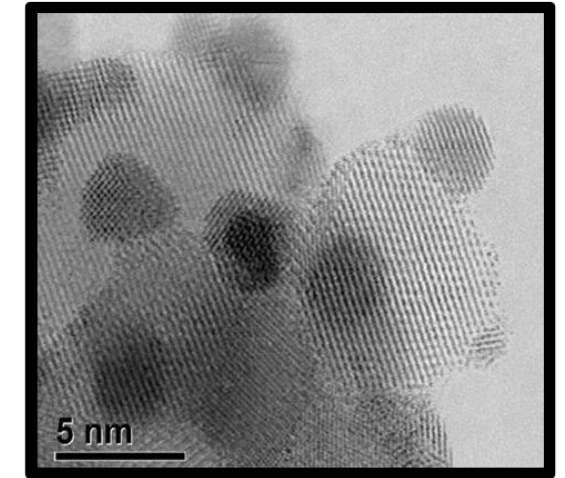
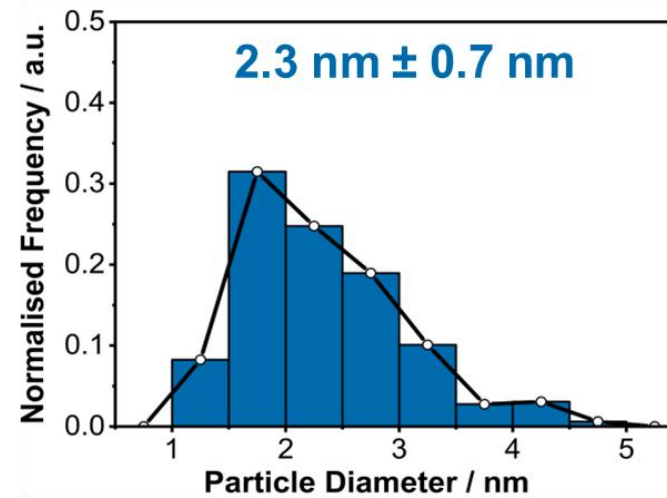
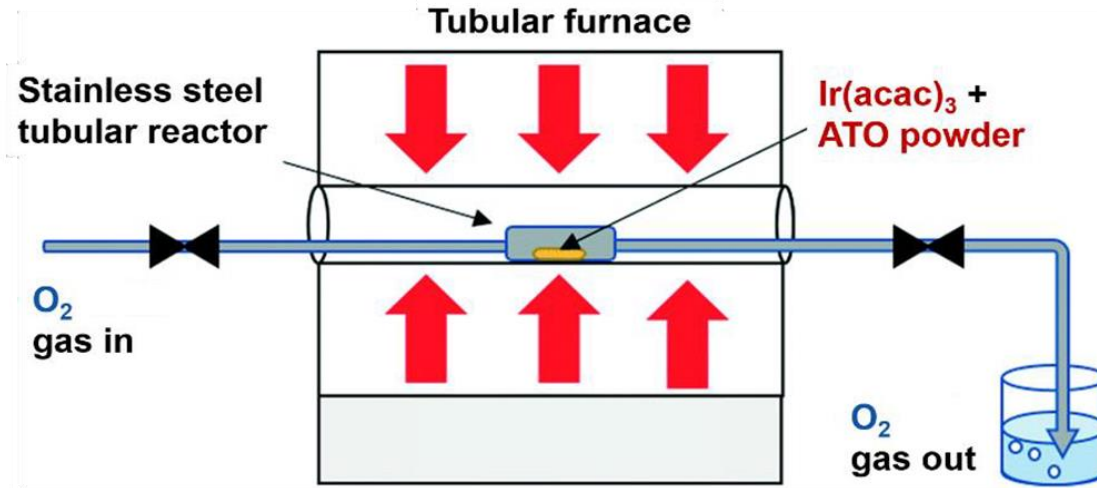
Polyol	Chemical Reduction	Adam's Fusion
		
Agglomerated particles that are a mixture of metallic Ir and IrO ₂ , ethylene glycol left behind	Highly agglomerated particles that are a mixture of metallic Ir and IrO ₂	Large particle sizes with a dominating presence of rutile IrO₂

- ❑ The development of **rutile iridium dioxide** nanoparticles that are **small** and **highly dispersed** has been a significant challenge in the field

IrO₂ Supported on ATO



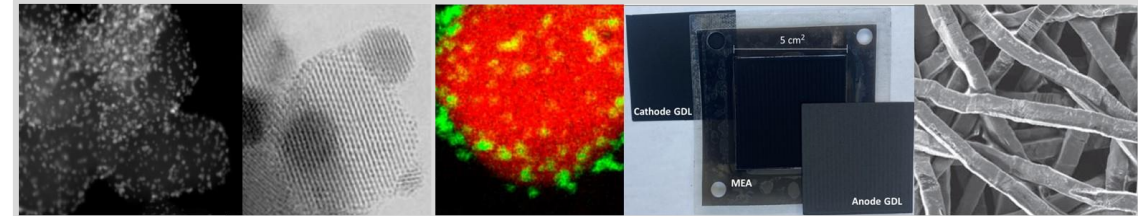
Small, highly dispersed rutile iridium dioxide nanoparticles



Evidence of a **structural metal-support interaction (SMSI)** in the form of **epitaxial particle anchorage**

Rajan, Z.S.H.S., et al. *Catalysis Science and Technology*, 2020.

IrO₂ Supported on ATO

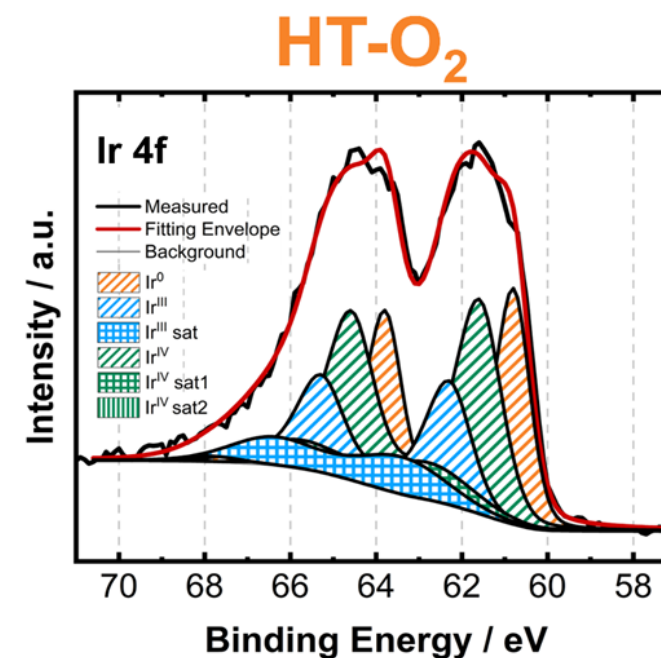
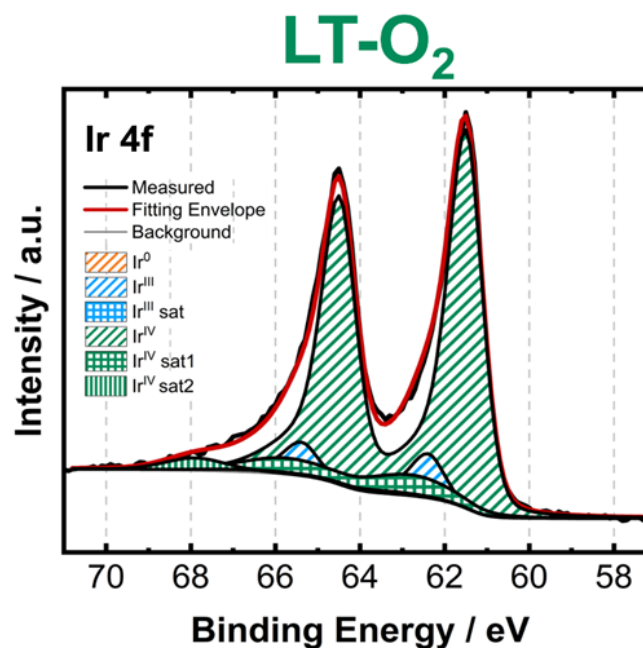
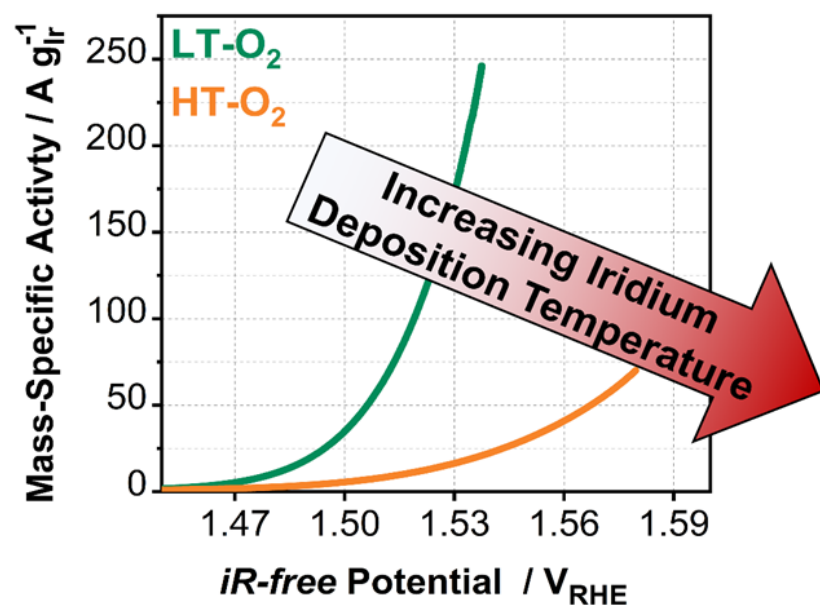


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Reductive Segregation During the Synthesis of Sb–SnO₂-Supported Iridium Electrocatalysts for the Oxygen Evolution Reaction

Ziba S. H. S. Rajan,¹ Tobias Binninger,² Patricia J. Kooyman,³ Darija Susac,¹ and Rhiyaad Mohamed^{1,z}



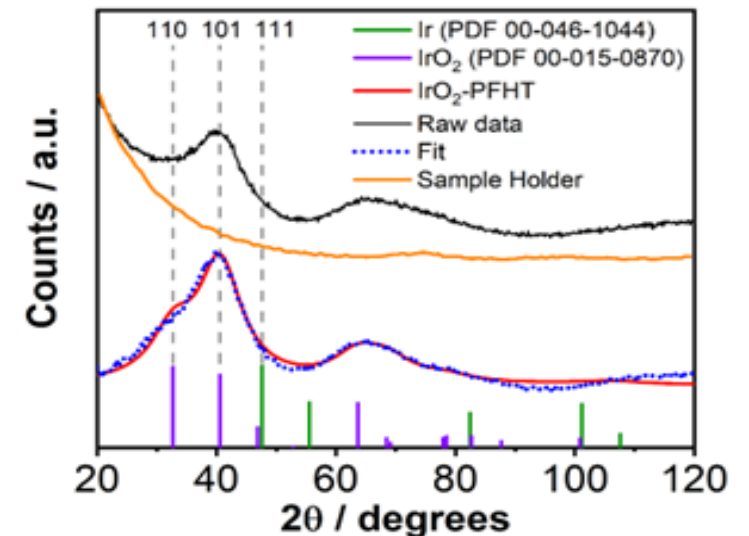
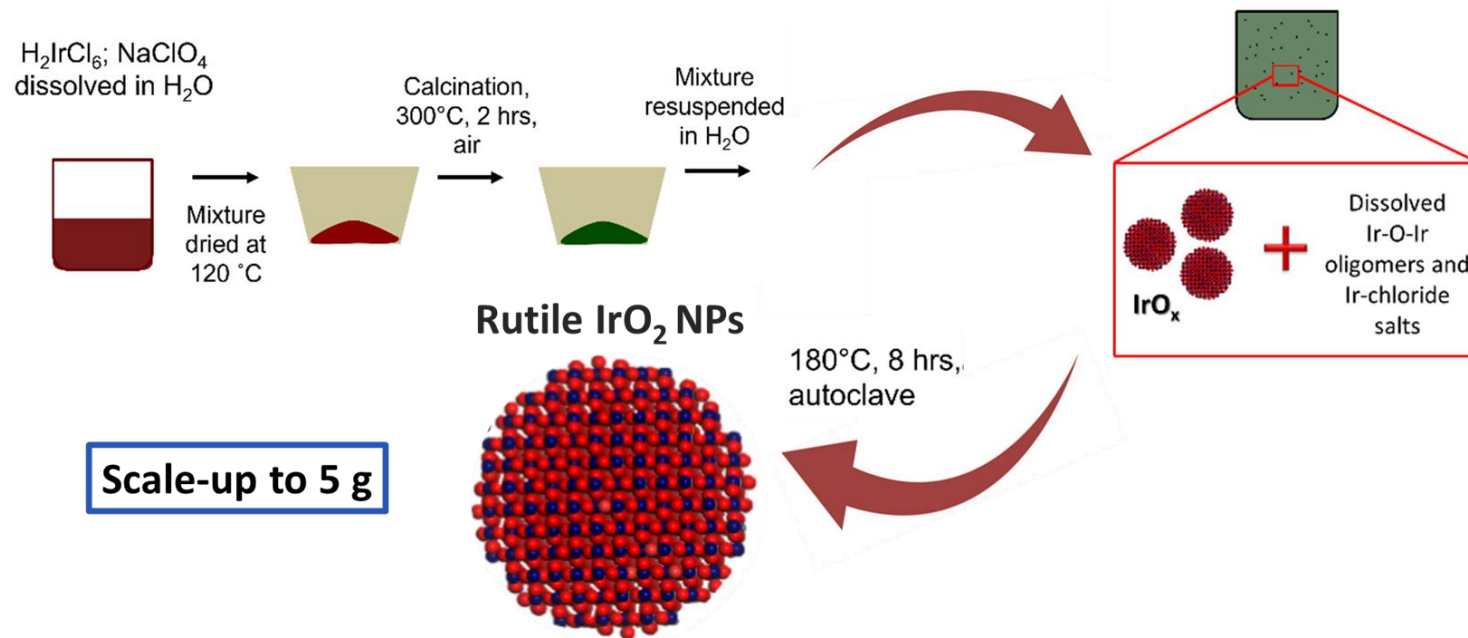
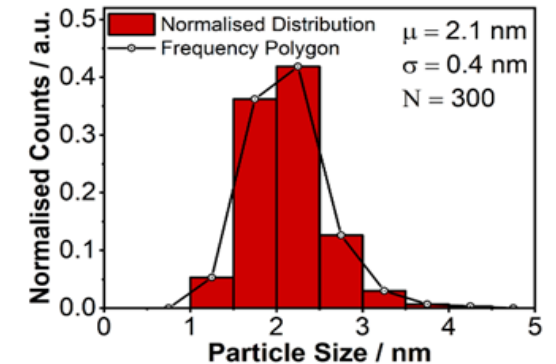
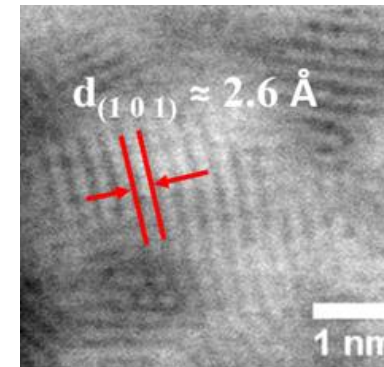
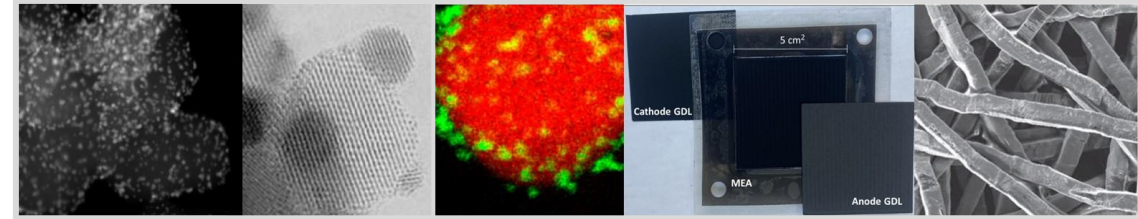
PFHT IrO₂

RESEARCH ARTICLE

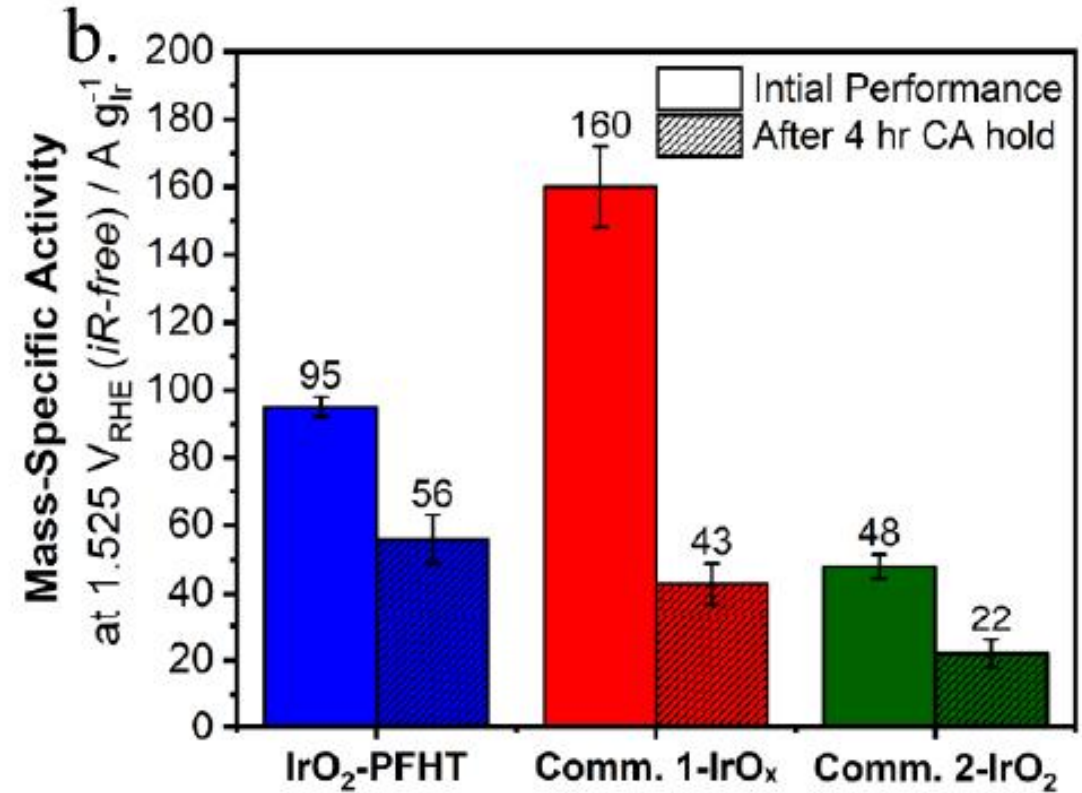
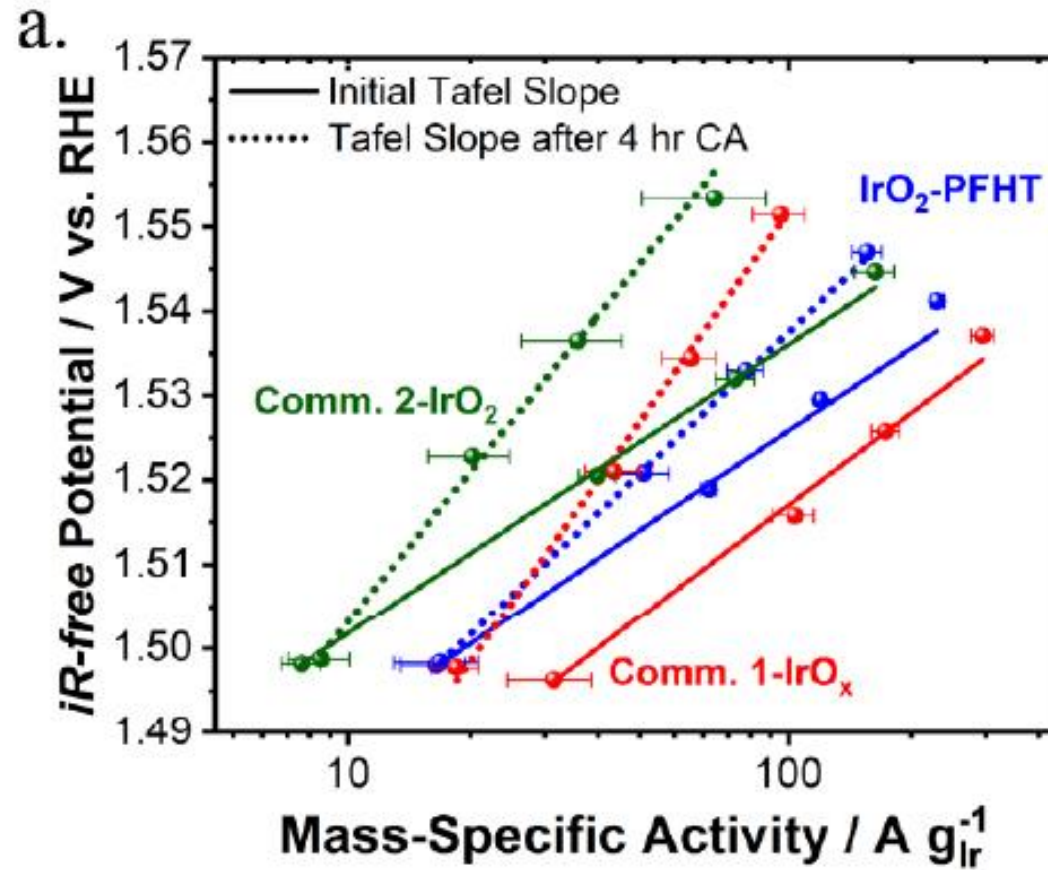
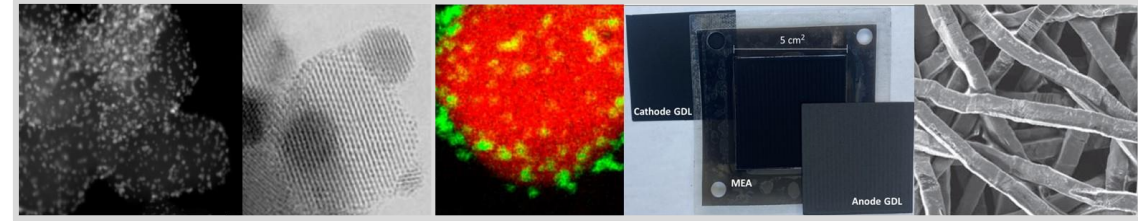
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Perchlorate Fusion–Hydrothermal Synthesis of Nano-Crystalline IrO₂: Leveraging Stability and Oxygen Evolution Activity

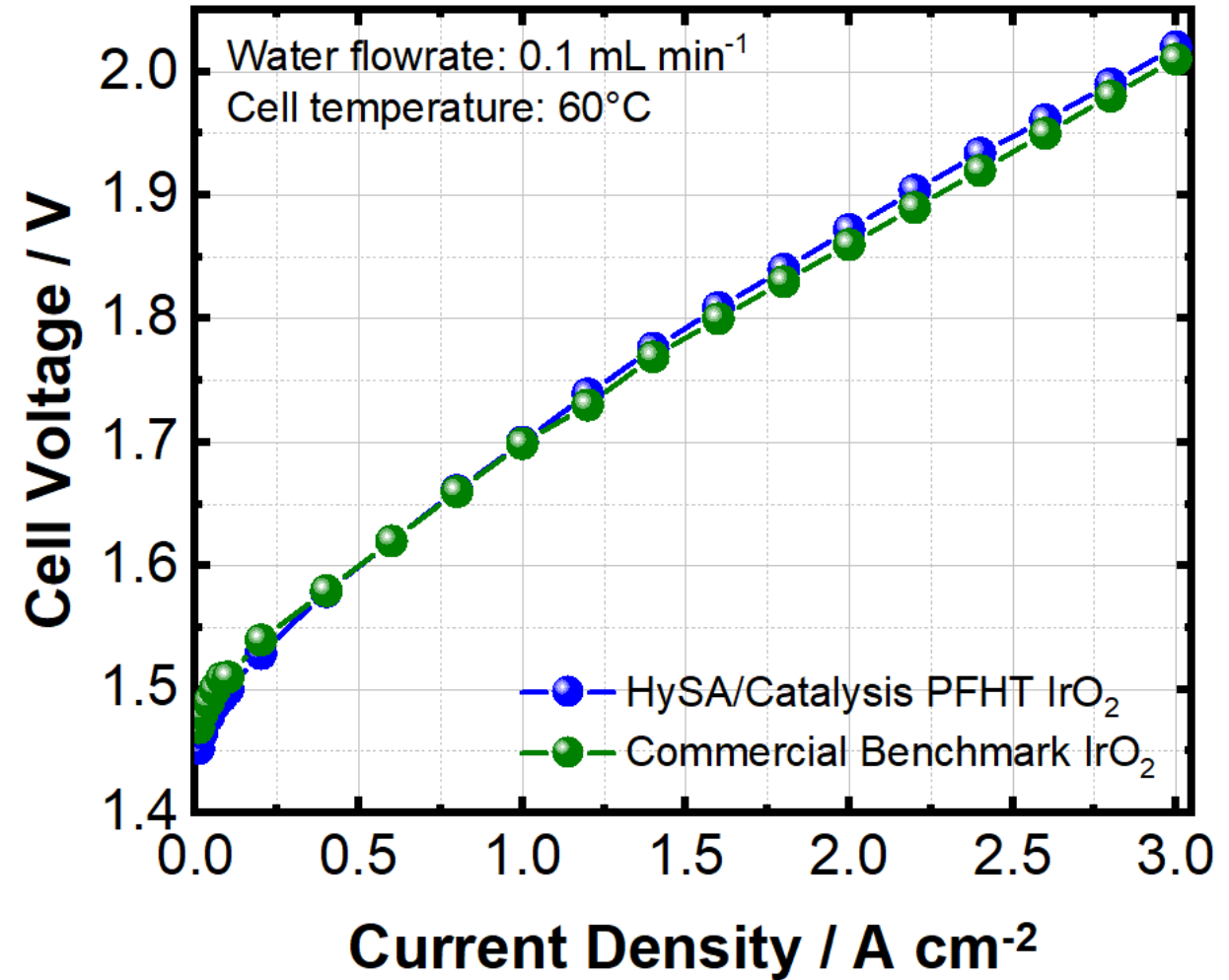
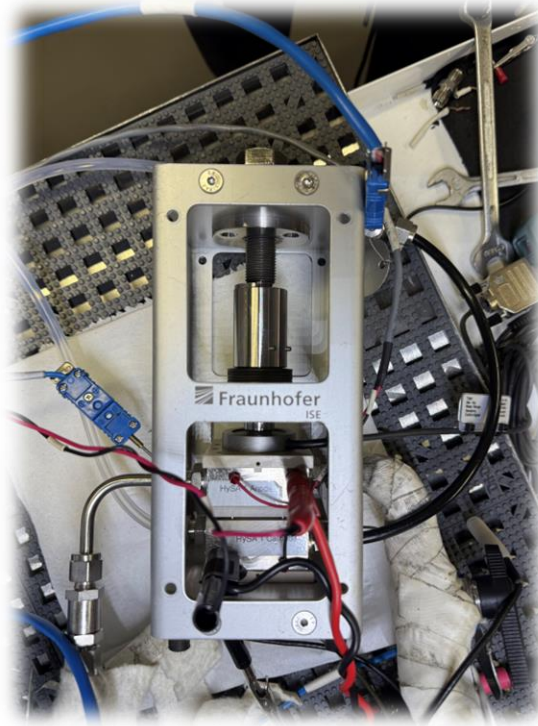
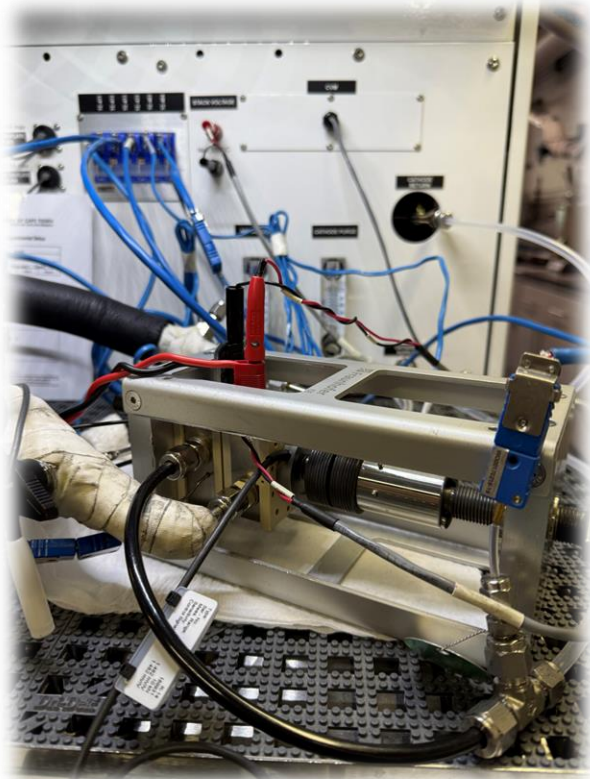
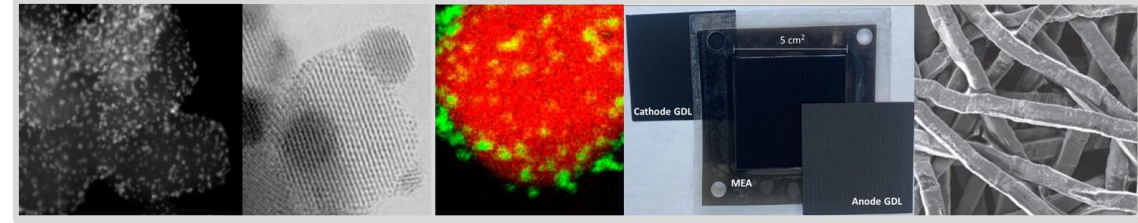
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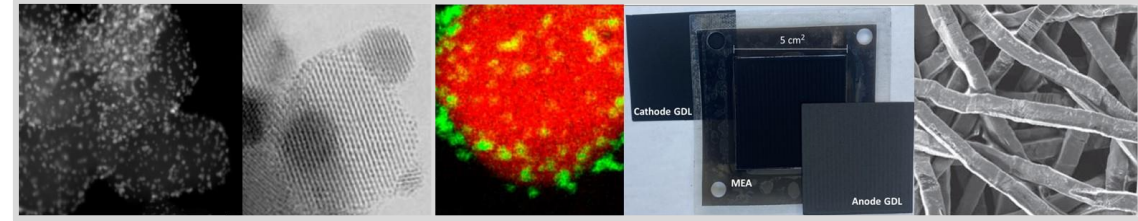
PFHT IrO₂ – *Ex-Situ* Performance



PFHT IrO₂ – *In-Situ* Performance

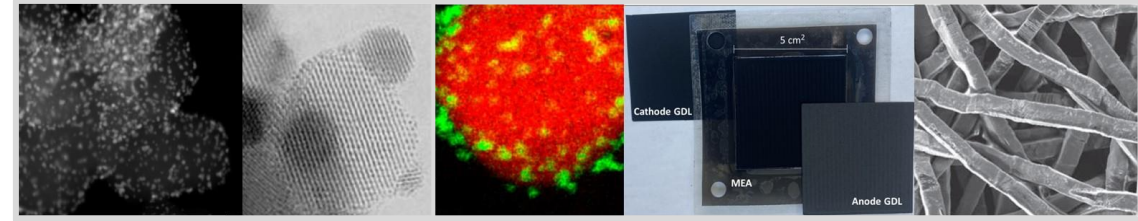


Conclusions and Future Work



- ❑ We have successfully prepared and characterised a **range of catalyst designs** consisting of **rutile-type IrO₂**
- ❑ **MOCD** has provided the **ideal basis for us to develop a fundamental understanding** towards the **behaviour of oxide-supported iridium oxide catalysts for the OER**
- ❑ The **IrO₂-PFHT** catalyst has demonstrated **favourable electrochemical performance in both *ex-situ* and *in-situ* conditions**
- ❑ Our future work consists of developing a **robust *in-situ* stability protocol**, composed of the **hundreds of hours online**, as well as the incorporation of our **supported catalyst range in a working electrolyser**

Thank you for your attention



science, technology
& innovation

Department:
Science, Technology and Innovation
REPUBLIC OF SOUTH AFRICA



BMBF



giz

Deutsche Gesellschaft
für Internationale
Zusammenarbeit (GIZ) GmbH

IMPLATS
Distinctly Platinum



<https://ebe.uct.ac.za/catalysis-institute>