



Determination of Rhodium using Fire Assay with Co-Collectors

Rampfumedzi Tshilidzi

*9th International PGM Conference
Sun City, Rustenburg, South Africa*

27 - 28 October 2025

OUTLINE

INTRODUCTION

OVERVIEW OF PGMs MARKET

**CHALLENGES ASSOCIATED
WITH DETERMINATION OF RH**

**METHODOLOGY AND
EXPERIMENTAL RESULTS**

CONCLUSION

ACKNOWLEDGEMENTS

❑ Background Overview of Rh

- ❑ Rhodium, one of six PGMs, is a silver-white metallic element that is highly reflective and strongly resistant to corrosion.
- ❑ Found mainly in platinum ores.
- ❑ Oxidation states: +2, +3, +4, +6
(Rh^{3+} most stable).



OVERVIEW OF PGMs MARKET

4

- Used in jewellery, electronics, catalysis, and industrial coatings.
- High value: often more expensive than Pt & Pd.

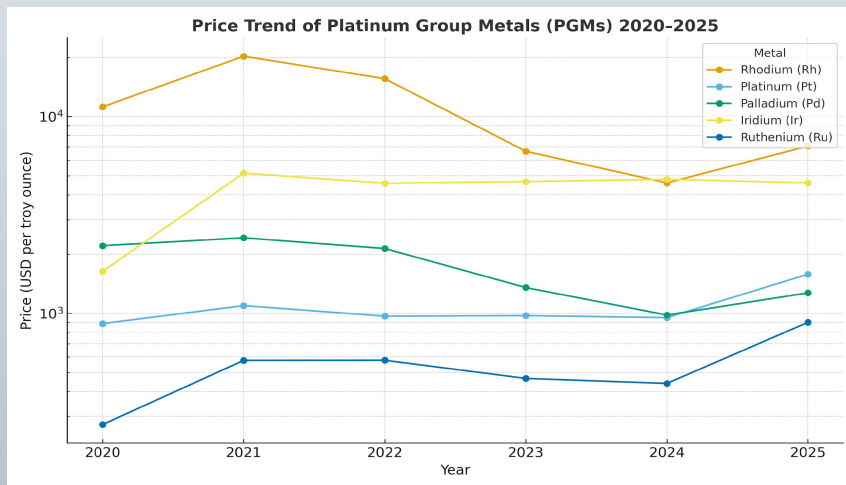
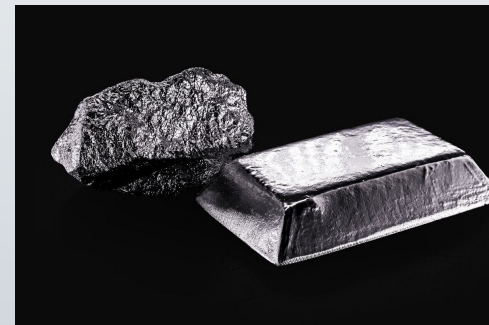


Figure 1. Price trend of Platinum Group Metals



□ Low Natural Abundance.

□ Spectral Interferences.

- *Rhodium emission lines (e.g., 343.489 nm, 369.236 nm, 384.787 nm) often overlap with strong lines from iron (Fe), platinum (Pt), and palladium (Pd).*

□ Matrix Effects

- *The presence of other PGMs and base metals (Ni, Cu, Co, Pb) in the assay matrix can suppress or enhance the Rh signal, affecting quantification.*

□ Sample Dissolution Issues

- *Rhodium is chemically inert and resistant to acid dissolution. Complete dissolution often requires high-temperature digestion.*

□ Interaction with Other PGMs

- *Rhodium recovery can be affected by the presence of other PGMs (Pt, Pd), as they can alter the metallurgical behavior of the prill.*
- *The addition of co-collectors such as Pd, Te, Cu, or Co can improve recovery, but the interaction must be optimized carefully.*

□ Analytical Variability

- *Due to these challenges, the precision and reproducibility of Rh results in fire assay can be poor compared to other PGMs.*
- *Variations in recovery often exceed $\pm 20\%$, particularly without co-collector optimization.*

□ Efficient Preconcentration

- *Fire assay ensures complete alloying of Rh with lead or nickel sulphide collectors.*
- *Fire assay concentrates Rh into a small metallic prill, enabling accurate detection.*

□ Matrix Separation in High-Temperature Recovery

- *Removes silicate and base metal matrices through slag formation.*
- *Produces a clean metallic phase suitable for further dissolution.*
- *The addition of co-collectors such as **Pd, Ag, Te, Cu, or Co** can improve recovery, but the interaction must be optimized carefully.*

- Investigate collection of Rh by fire assay with lead flux + co-collectors

(Silver (Ag), Palladium (Pd), Tellurium (Te)).

- Investigate acid dissolution methods with the open digestion method.
- Evaluate analytical sensitivity and recovery using ICP-OES

FIRE ASSAY FUSION

950–1200 °C

Pb flux + co-collectors



Lead prill

- litharge (PbO, 54.50%)
- sodium carbonate (Na_2CO_3 , 27.00%)
- silica powder 325 (SiO_2 , 4.50%)
- borax anhydrous ($\text{Na}_2\text{B}_4\text{O}_7$, 13.00%)
- kerosol 150/200 (1.00%)

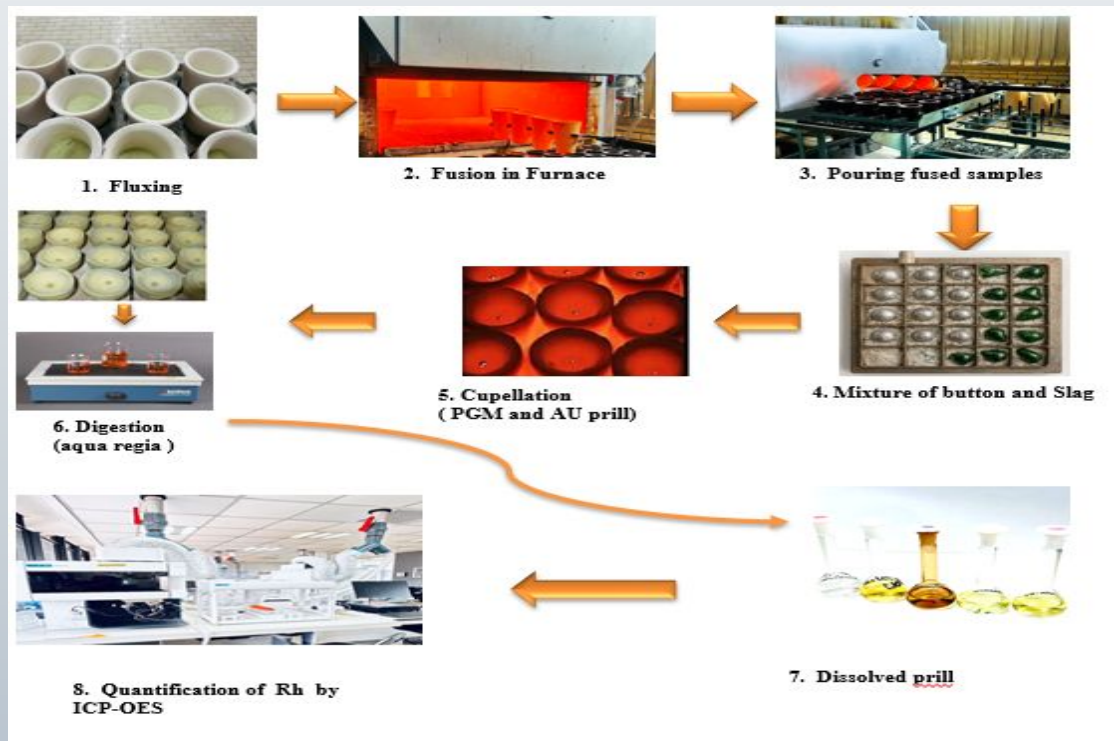
CERTIFIED REFERENCE MATERIALS

SARM 107

UG2 chromitite ore concentrate
Rh = 0.32 ppm
low-grade CRM

AMIS 314

Merensky Reef flotation concentrate
Rh = 1.89 ppm
high-grade CRM



Cupellation Process (Scientific Expression)

Main Reaction: $\text{Pb (s)} + \frac{1}{2} \text{O}_2 \text{ (g)} \rightarrow \text{PbO (l/s)}$

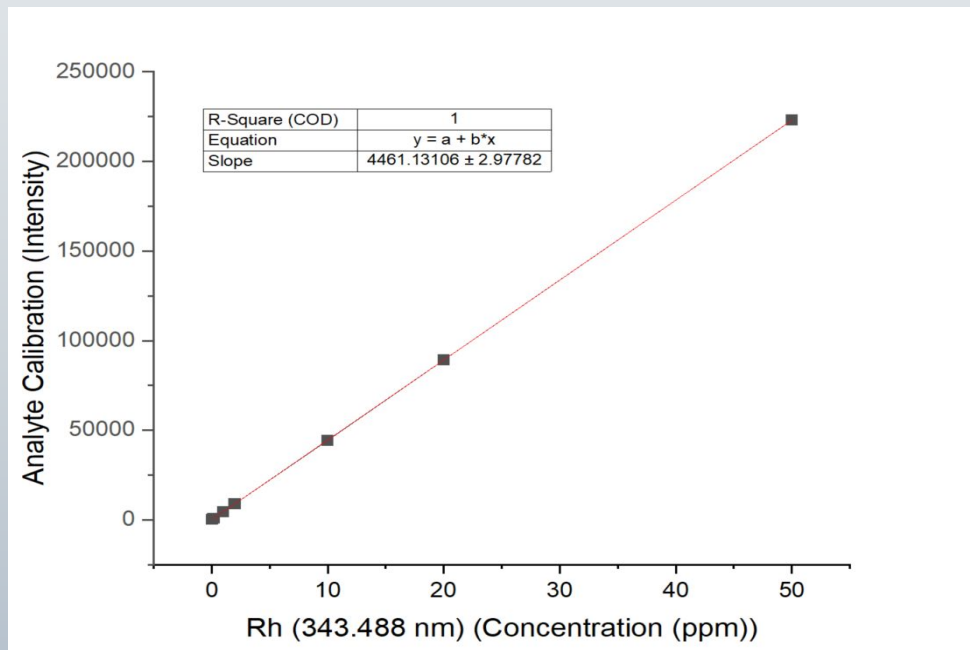
Absorption Process: $\text{PbO (l)} \rightarrow \text{PbO (absorbed in cupel)}$

- As cupellation proceeds, litharge (PbO) is continuously absorbed by the magnesia or bone-ash cupel, effectively removing lead and other base metal oxides.

Final Result: $\text{Au (s)} + \text{PGMs (s)} \rightarrow \text{Metallic bead residue}$

Figure 2. Illustration of fire assay process for Rh determination

□ ICP-OES was used to provided reliable quantitative results



- ICP-OES analysis with calibration 0.05–50 ppm.

Figure 3. Analytical calibration from ICP-OES

Summary of Rh recovery from SARM 107

Sample ID	Concentration (ppm)			
	Spike (2ml)			
	No spike	Ag	Pd	Te
SARM 107/1	0.05	0.01	0.27	0.26
SARM 107/1	0.12	0.00	0.25	0.26
SARM 107/1	0.12	0.00	0.25	0.32
SARM 107/1	0.20	0.02	0.26	0.27
Average Recovery	0.12	0.01	0.26	0.28
Accurate Value	0.32	0.32	0.32	0.32
SD	0.06	0.01	0.01	0.03
RSD	52.47	106.07	3.94	10.78
Recovery %	37,5	3.20	81.25	87.50

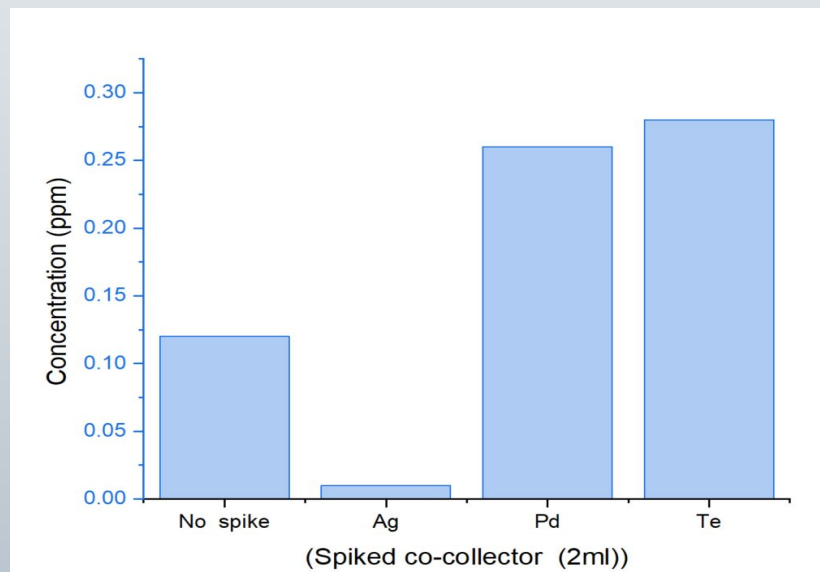


Figure 4. Average recorded concentration of Rh

Summary of Rh recovery from AMIS 314

Sample ID	Concentration (ppm)			
	Spike (2ml)			
	No spike	Ag	Pd	Te
AMIS 314/1	1.12	0.03	1.77	1.63
AMIS 314/1	1.40	0.02	1.90	1.71
AMIS 314/1	1.68	0.02	1.71	1.61
AMIS 314/1	1.20	0.02	1.45	1.81
Average Recovery	1.35	0.02	1.71	1.69
Accurate Value	1.89	1.89	1.89	1.89
SD	0.25	0.00	0.19	0.09
RSD	18.44	16.66	11.01	5.43
Recovery %	71.42	1.05	90.48	90.00

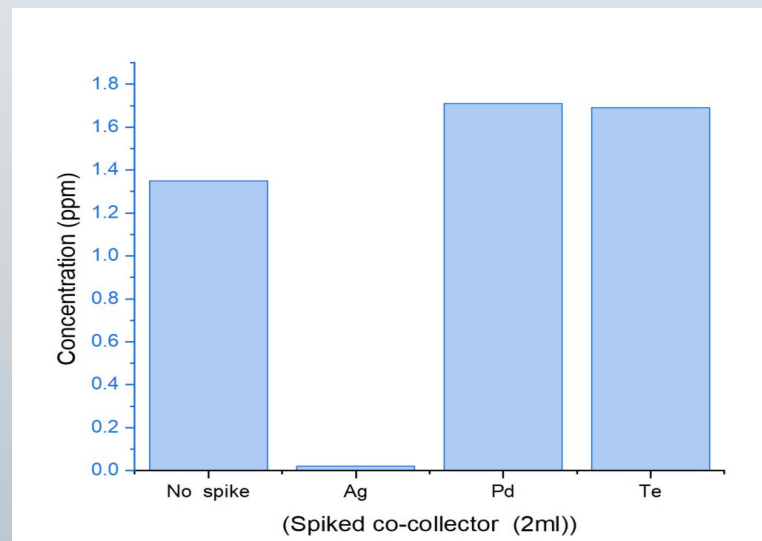


Figure 5. Average recorded concentration of Rh

CONCLUSION

- Determination of Rh requires tailored analytical approaches.
- Co-collectors improve PGM collection and analytical sensitivity depending on interactions.
- Palladium (Pd) and Tellurium (Te) are most effective for Rh, good recovery (100–120%).
- Fire assay with lead collection + co-collectors improves preconcentration of PGMs.
- ICP-OES provided reliable quantitative results.
- Additionally, other dissolution methods such as high-pressure or closed-vessel microwave digestion can be employed to enhance recovery.

- ❑ Mintek
 - Analytical Chemistry Division team
- ❑ Southern African Institute of Mining And Metallurgy (SAIMM)



THANK YOU

FOR YOUR ATTENTION

Questions and Discussion Welcome

200 Malibongwe Drive
Randburg
South Africa

Tel: +27 11 709 4111

www.mintek.co.za

MINTEK

Private Bag X3015
Randburg
2125
South Africa



Mintek



MintekSA



Mintek_RSA



Mintek_RSA