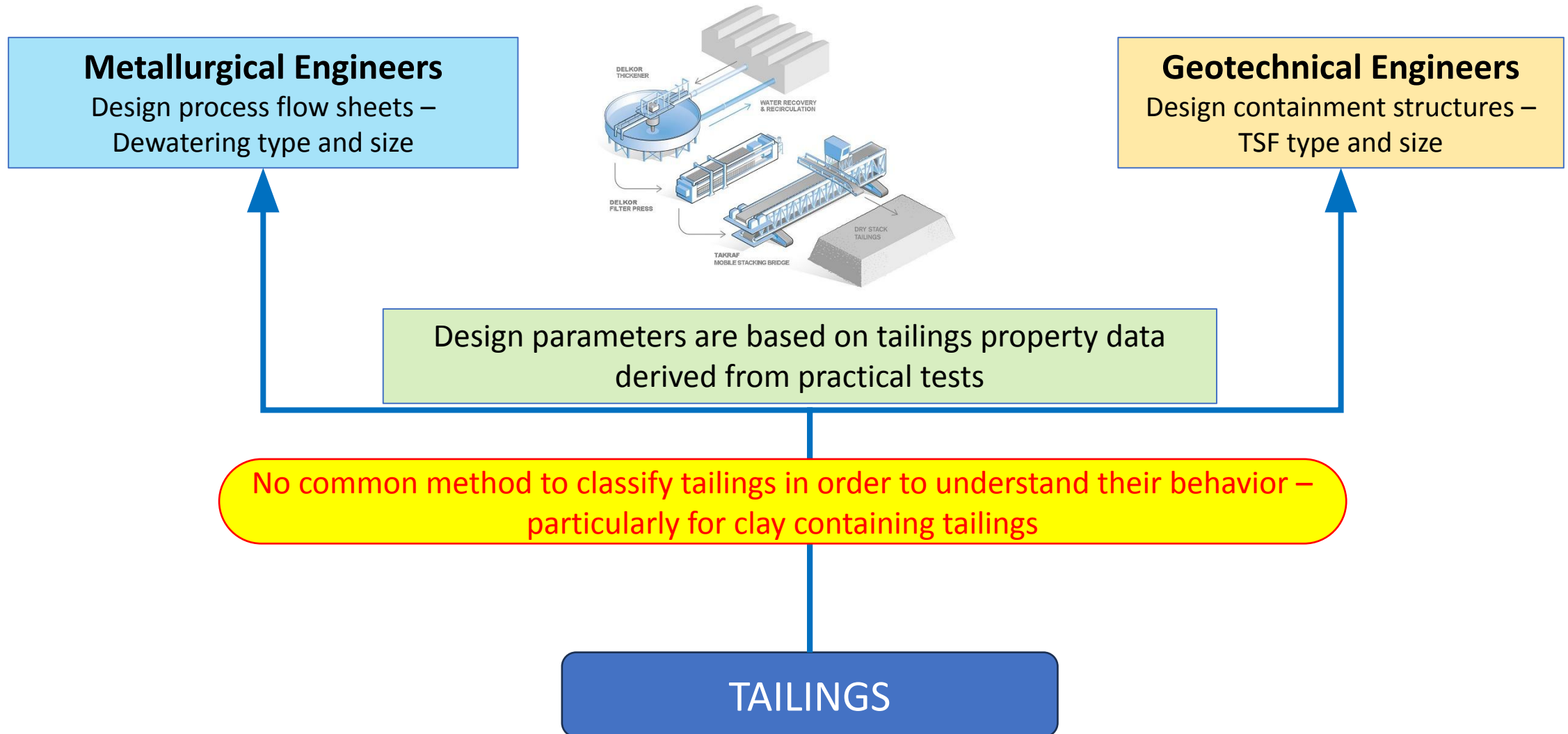


Classification of tailings by clay minerals and pore water chemistry as a basis for safe tailings disposal

A.J. Vietti and N.C. Steenkamp

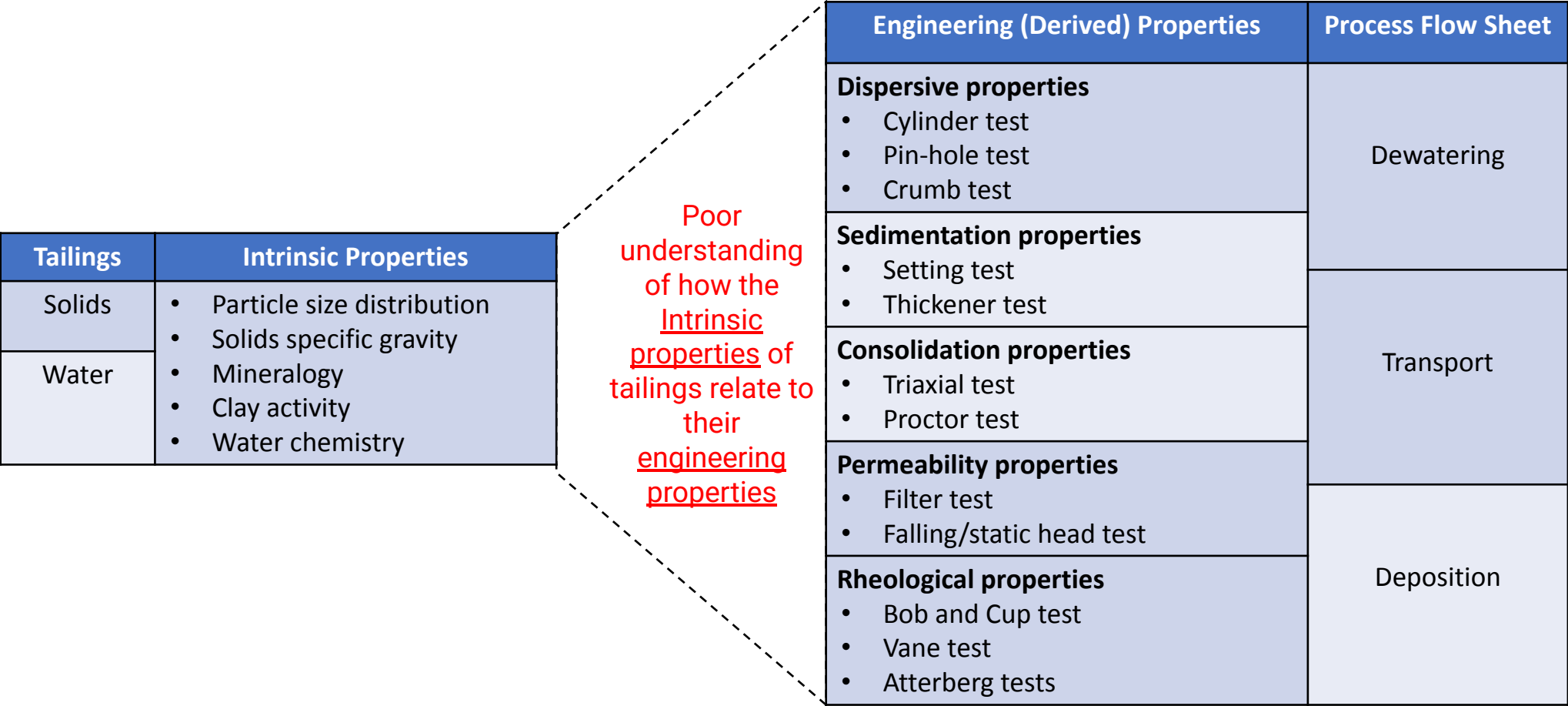


Tailings Properties Overview



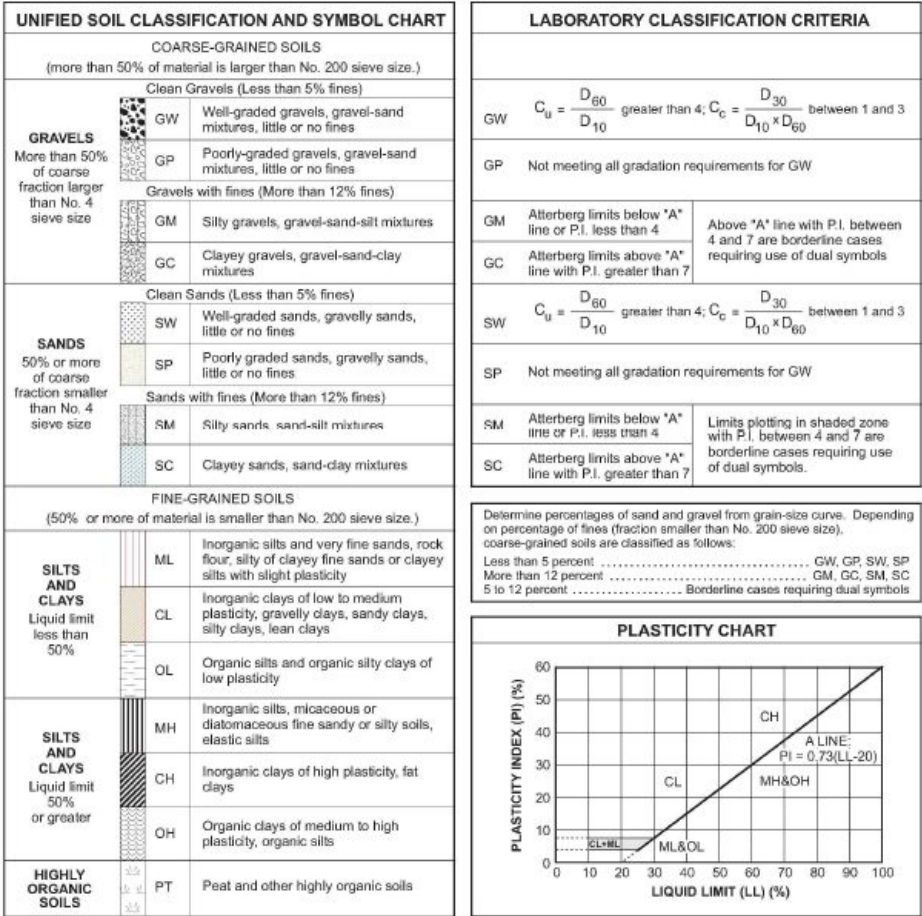
Tailings Properties Overview

- Current framework for classifying tailings



Current Tailings Classification Method

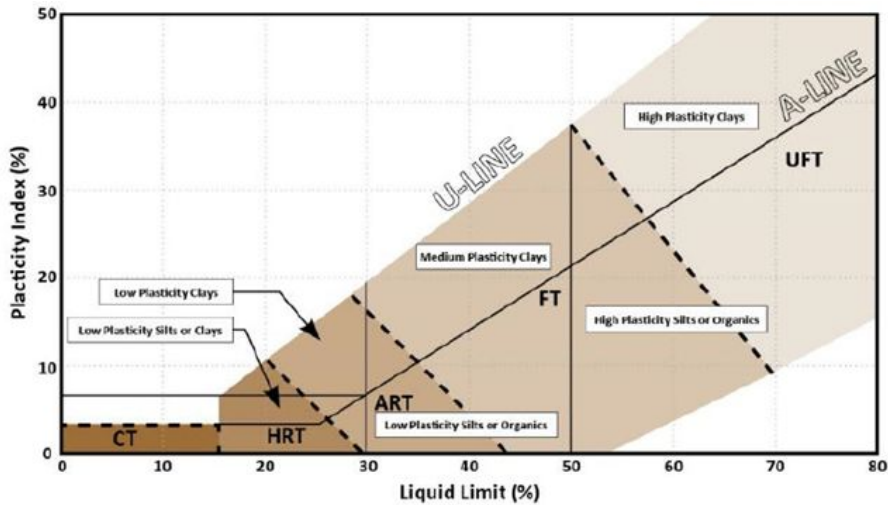
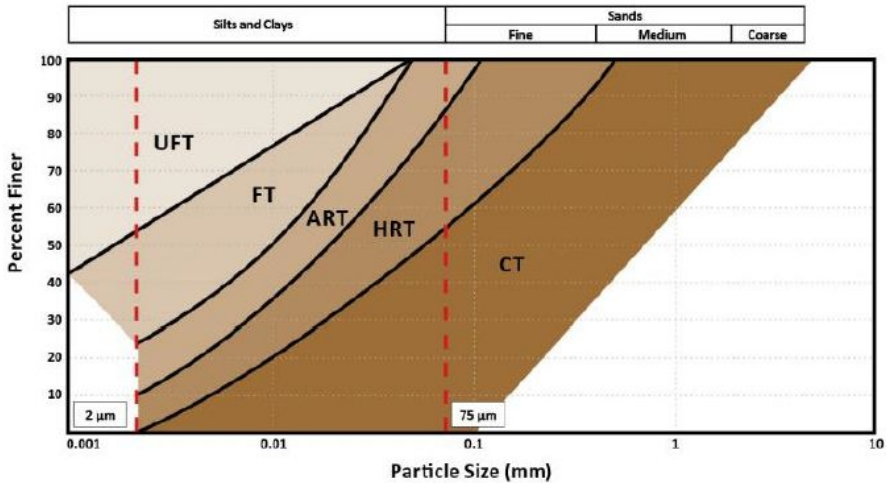
- Geotechnical Engineers commonly use the Unified Soil Classification System (USCS) for soils.
- Classify by particle size (intrinsic property) and by Atterberg limits (water quality dependent) into cohesive and non-cohesive groups



Current Tailings Classification Method

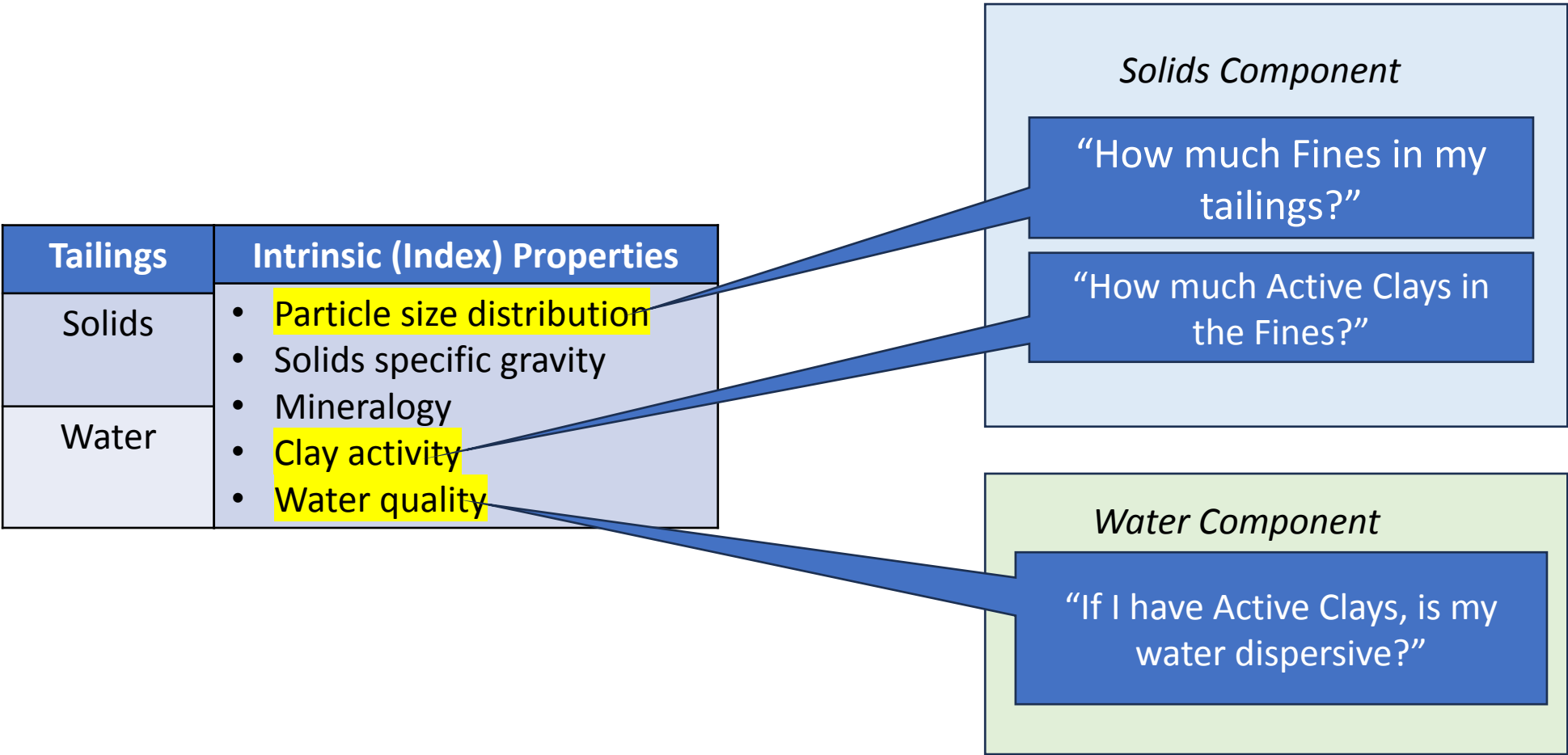
- Recently (2021) ICOLD introduced a “Tailings Dam Design Technology Update” suggesting tailings be classified into 5 groups.
- Provide “property templates” to classify tailings

Tailings Type	Symbol	Description (compare)	Example of mineral/ore
Coarse tailings	CT	Silty SAND, non-plastic	Salt, mineral sands, coarse coal rejects, iron ore sands
Hard Rock tailings	HRT	Sandy SILT, non to low plasticity	Copper, massive sulphide, nickel, gold
Altered Rock tailings	ART	Sandy SILT, trace of clay, low plasticity, bentonitic clay content	Porphyry copper with hydrothermal alteration, oxidized rock, bauxite, leaching processes
Fine tailings	FT	SILT, with trace to some clay, low to moderate plasticity	Iron ore fines, bauxite (red mud), fine coal rejects, leaching processes, metamorphosed/weathered polymetallic ores
Ultra Fine tailings	UFT	Silty CLAY, high plasticity, very low density and hydraulic conductivity	Oil sands (fluid fine tailings), phosphate fines; some kimberlite and coal fines



Current Tailings Classification Method

- Important questions to consider when classifying clay containing (cohesive) tailings

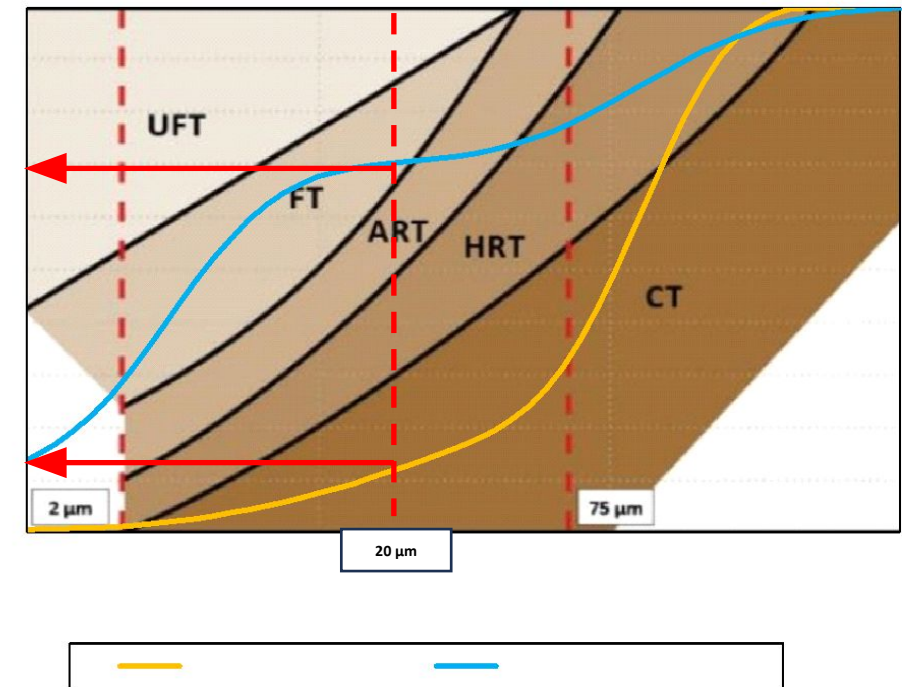
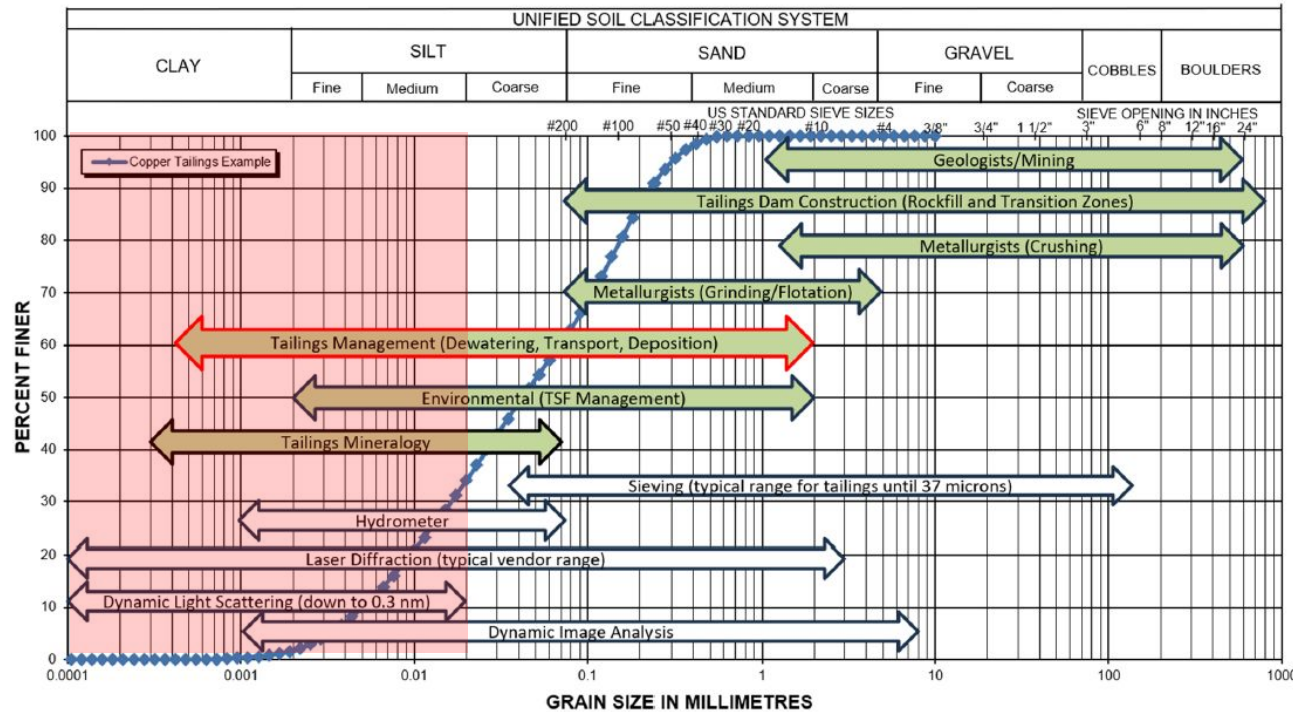


Classifying Tailings: Solids Component

- How much Fines do I have in my tailings?

Solids Particle Size

- Define -20 μ m as “Fines” fraction (rather than -2 μ m as “Clay”)
- Accepted as a valid criteria for classifying tailings behavior for dewatering and backfill applications



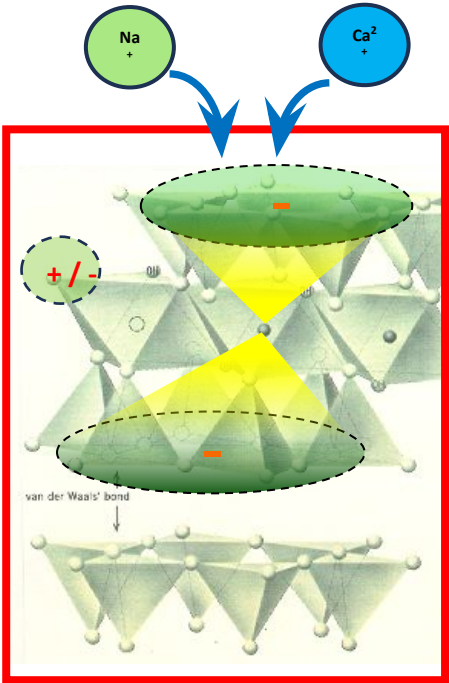
Engels, J, Guzman, J & Valdes, S 2025, 'Tailings particle size distribution methodology: the good, the bad, and the ugly', in AB Fourie, A Copeland, V Daigle & C MacRobert (eds), *Paste 2025: Proceedings of the 27th International Conference on Paste, Thickened and Filtered Tailings*, Australian Centre for Geomechanics, Perth, pp. 165-180,

Classifying Tailings: Solids Component

- How much Active Clays do I have in my Fines?

Active Clays have charged surfaces

- EDGES* – negative or positive charge depending on pH when in water.
- SURFACE* – negative which can exchange cations when in water.



Type	Charge on layer	Diocahedral	Trioctahedral
1:1	(x ~ 0)	Kaolinite kaolinite, dickite, nacrite, halloysite	Serpentine amesite, chrysotile, antigorite, lizardite
	(x ~ 0)	Pyrophyllite	Talc
2:1	(x ~ 0.2 – 0.6)	Smectite montmorillonite, beidellite, nontronite	Smectite saponite, hectorite, stevensite
	(x ~ 0.6 – 0.9)	Vermiculite	Vermiculite
	(x ~ 0.75 – 0.9)	Illite Glauconite	
	(x ~ 1.0)	Micas muscovite, paragonite, phengite, celadonite	Micas phlogopite, biotite, lepidolite
	(x ~ 2.0)	Brittle micas margarite, clintonite	
	(x ~ variable)	Chlorites donbassite	Chlorites clinochlore, chamosite, ripidolite
	(x ~ variable)		Sepiolites, Palygorskites
	(x ~ variable)		

Active/Swelling Clays

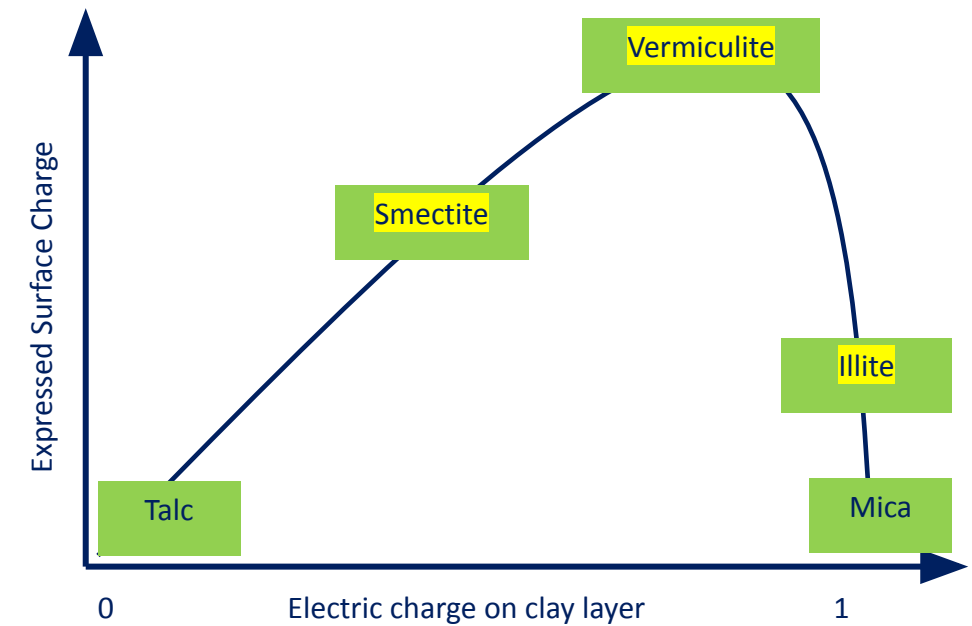
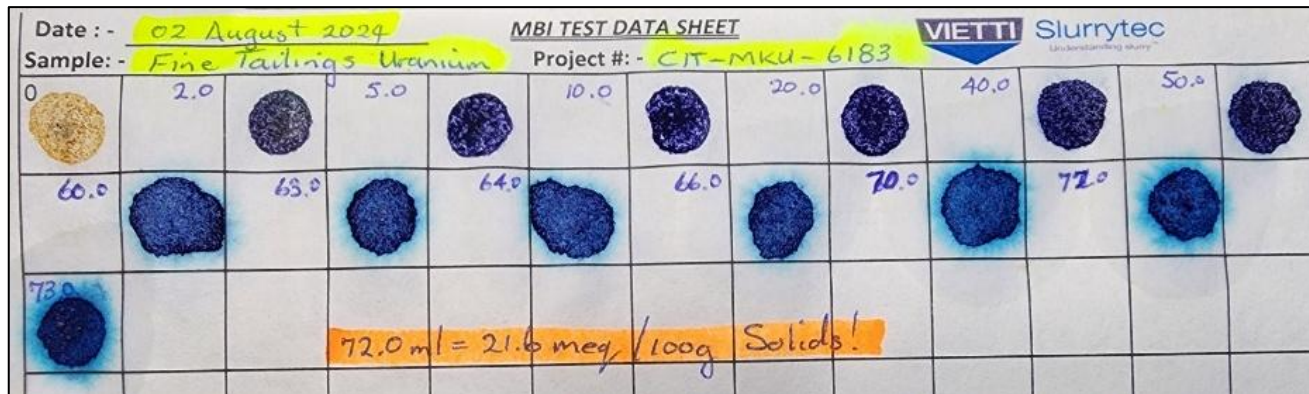
Grimm, RE, 1968, ‘Clay mineralogy (2nd ed)’, McGraw Hill, New York.

Classifying Tailings: Solids Component

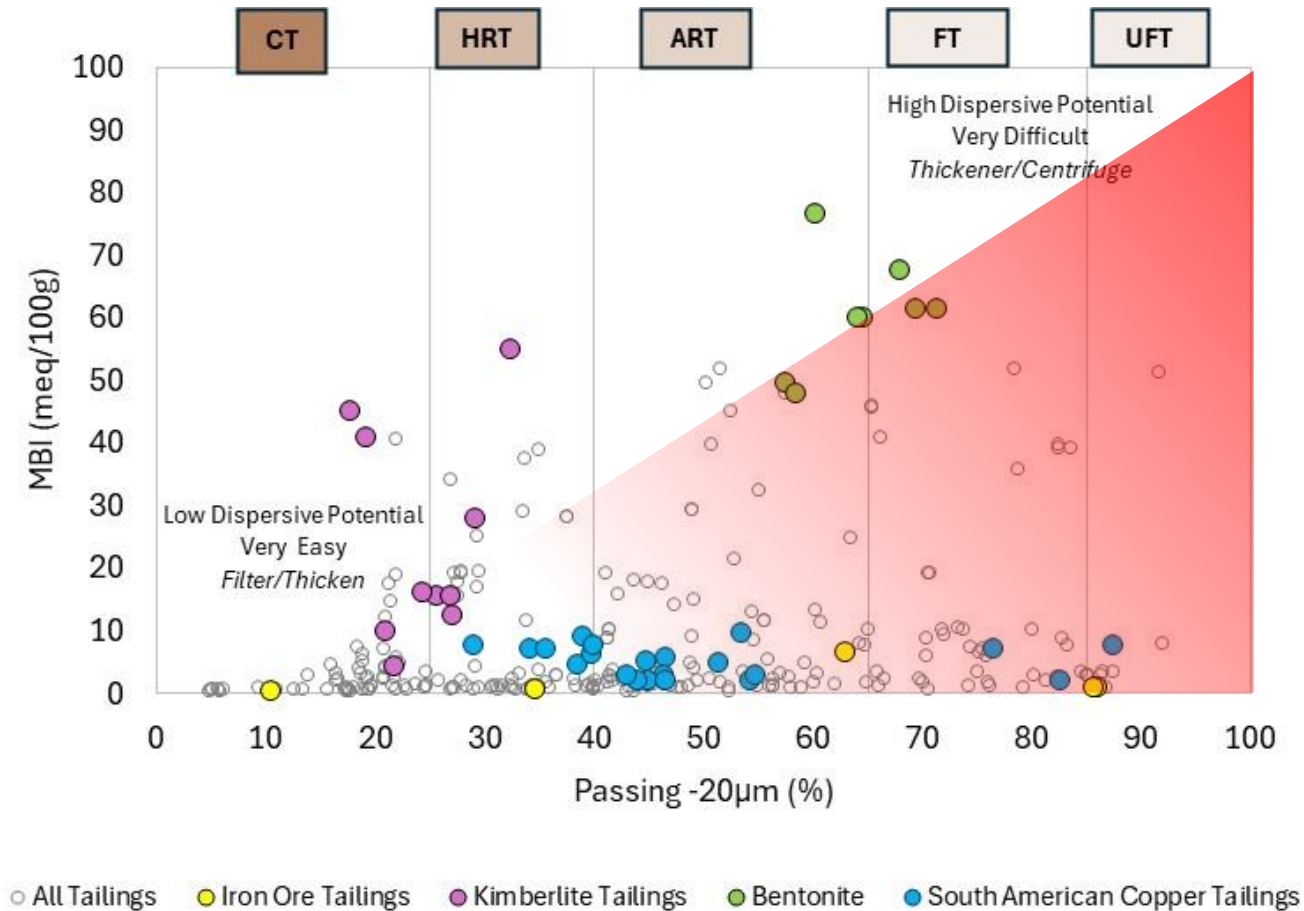
- How much Active Clays do I have in my Fines?

Clay Activity (expressed surface charge)

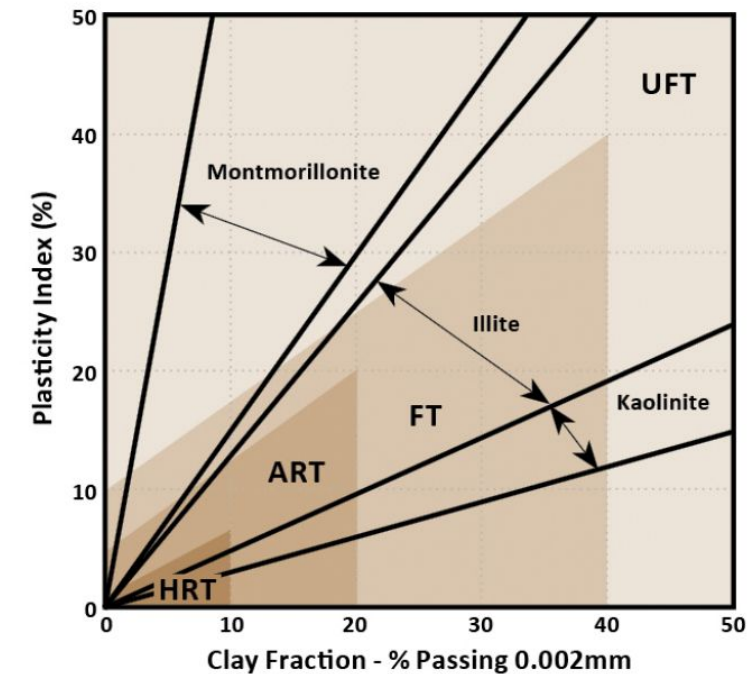
- Methylene Blue Index Method - Dye titration method to determine clay surface charge/area
- Standard Test Method for Methylene Blue Index of Clay (ASTM Standard C837)
- Define <15 meq/100g threshold as lower limit of Active/Swelling clay



Classifying Tailings: Solids Component



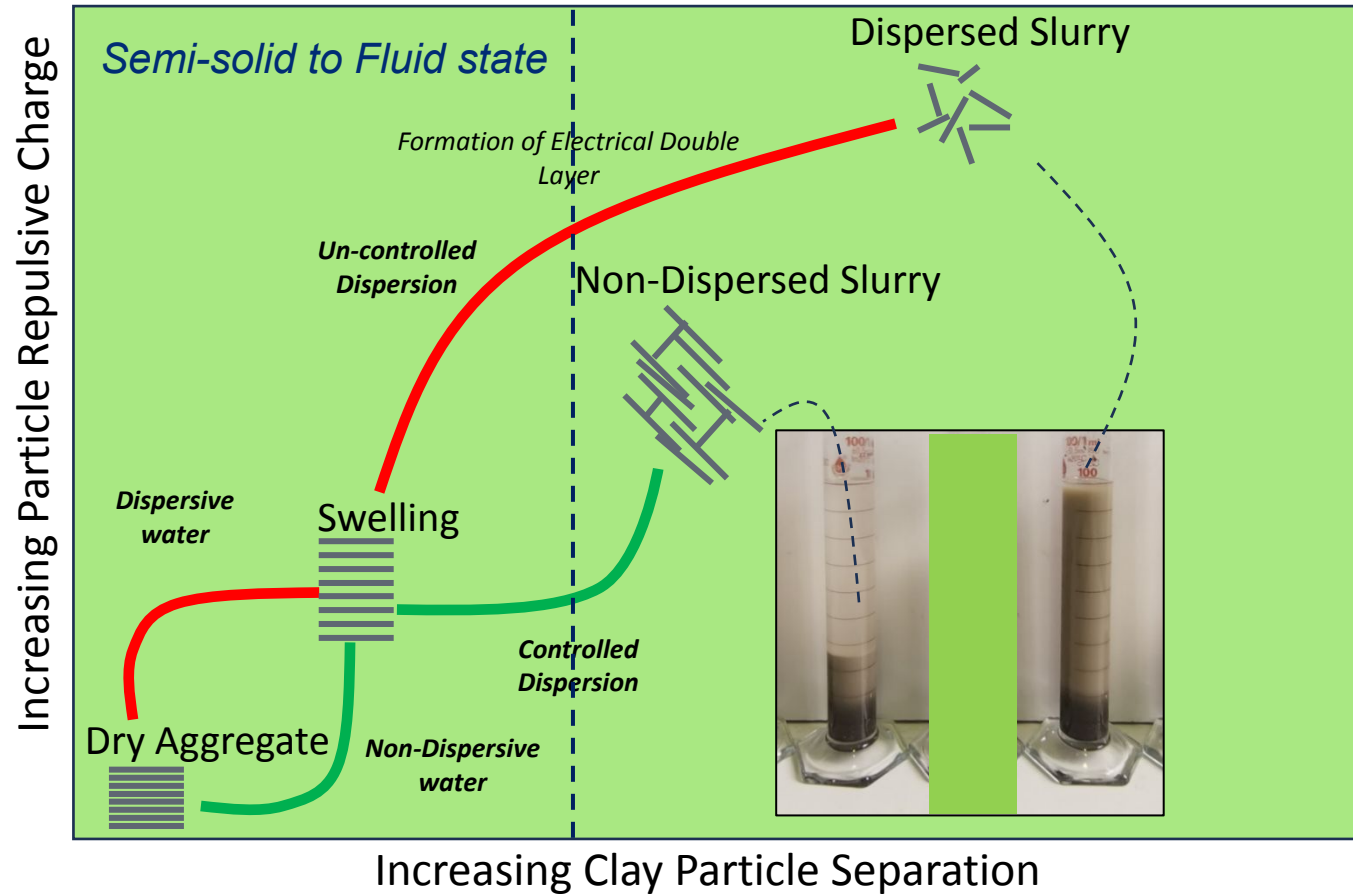
Similar to the Skempton Activity Chart



The Active Clay dispersive properties depend on the contacting water quality

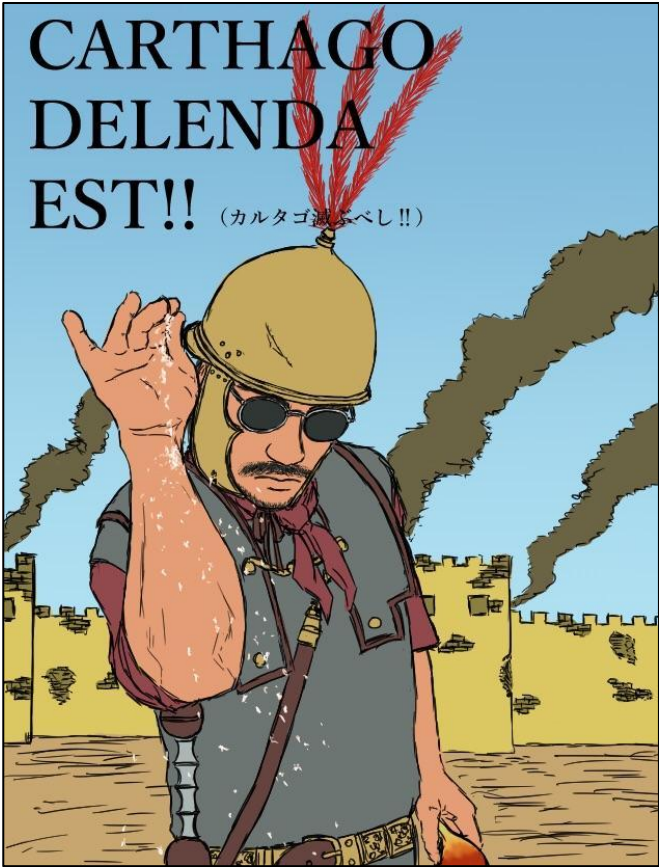
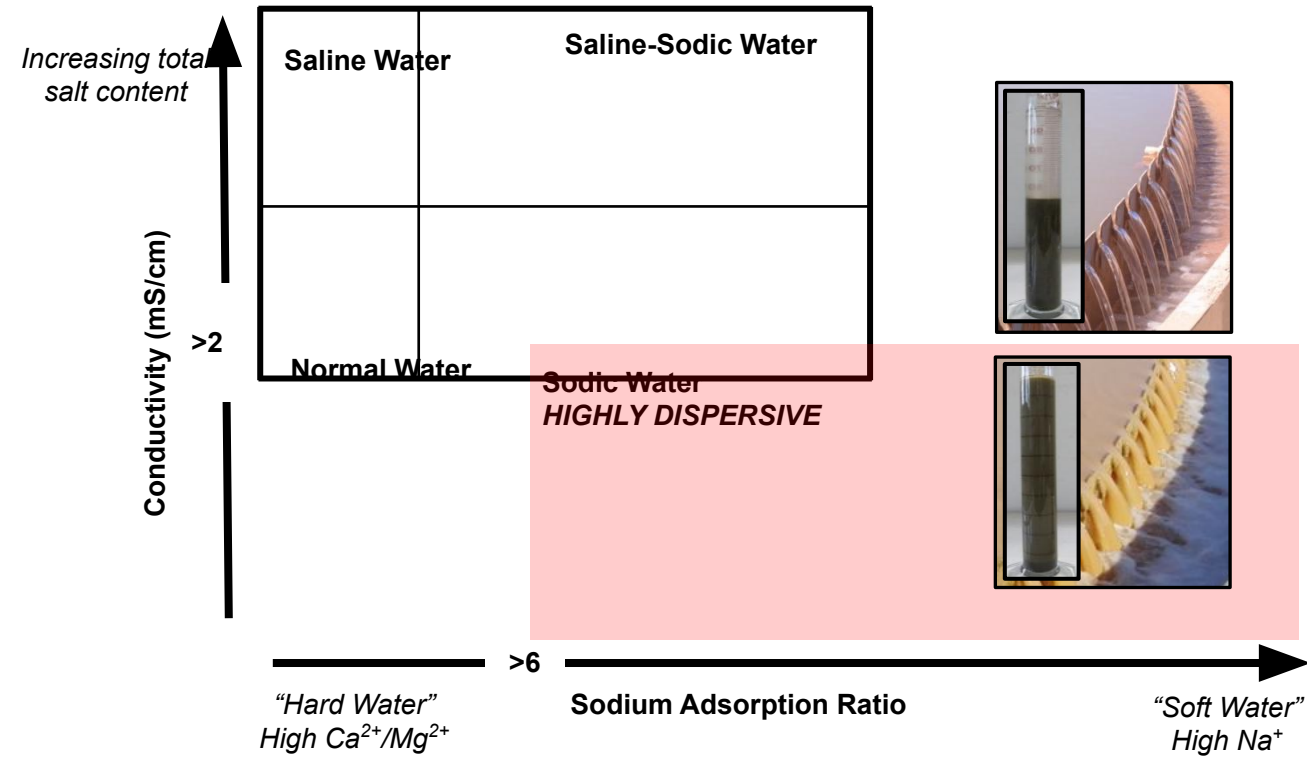
Classifying Tailings: Water Component

- If I have Active Clays, is my water dispersive?



Classifying Tailings: Water Component

- Tailings containing Active Clays have similar DISPERSIVE properties as Saline/Sodic soils



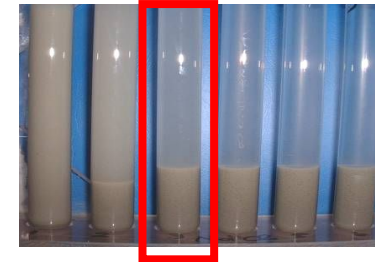
Fitzpatrick, RW, Boucher SC, Naidu R and Fritsch E 1994, 'Environmental consequences of soil sodicity' *Australian Journal of Soil Research*, vol 32, pp 1068-1093.

Classifying Tailings: Water Component

1. Conductivity of Water (mS/cm)

- Measured by conductivity meter
- Units defined in mS/cm
- Define <2.0 mS/cm as TEC for dispersion

Threshold Electrolyte Concentration(TEC)

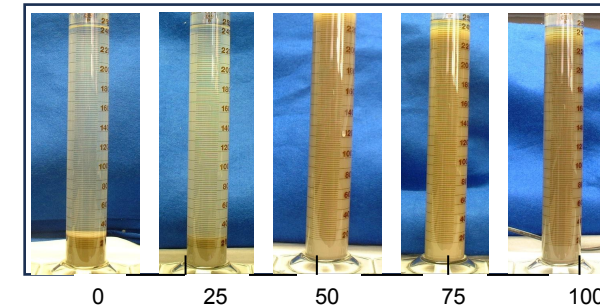


Increasing Conductivity

2. Sodium Adsorption Ratio of Water

- Water Quality Analysis
- Define >6.0 as threshold for dispersion
- Equation (in meq/l):

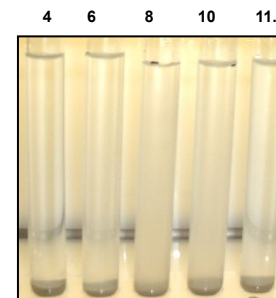
$$SAR = \frac{Na^+}{\sqrt{\frac{Ca^{2+} + Mg^{2+}}{2}}}$$



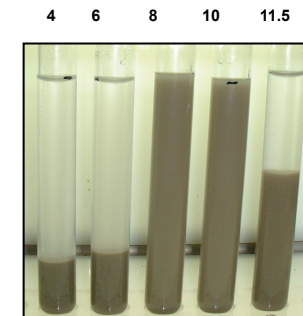
Increasing SAR

3. pH of Water

- Measured using a pH meter
- Define 8 to 10 for dispersion at >6 SAR



SAR <6



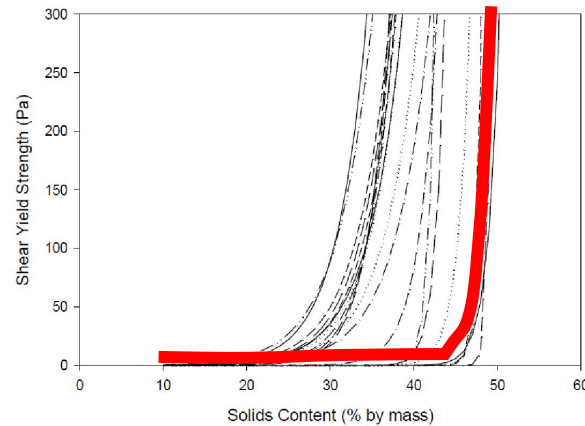
SAR >6

Classifying Tailings: Water Component

Dispersive slurry



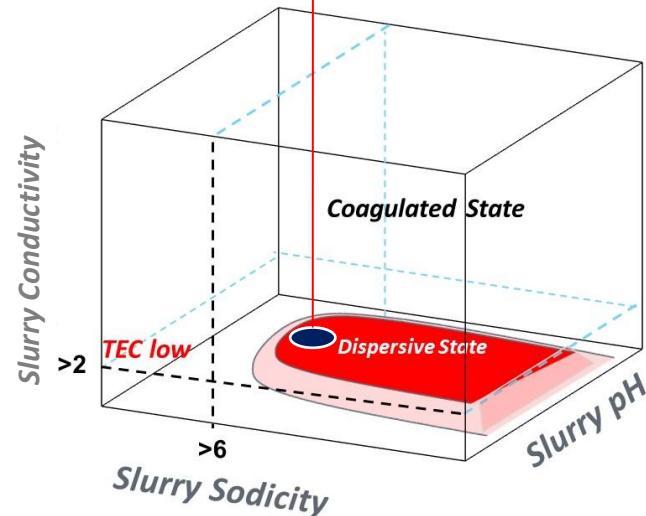
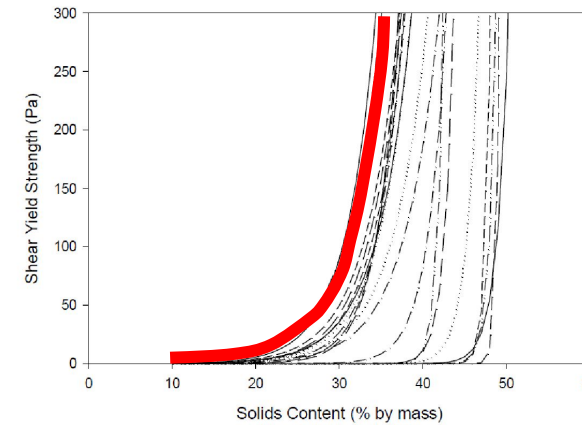
"Hockey stick shape"
Rheological property



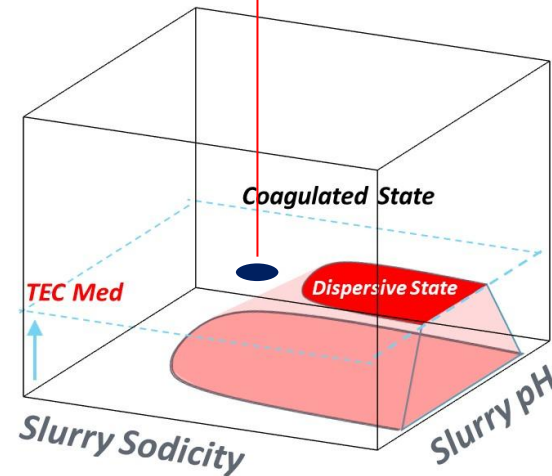
Non-dispersive slurry



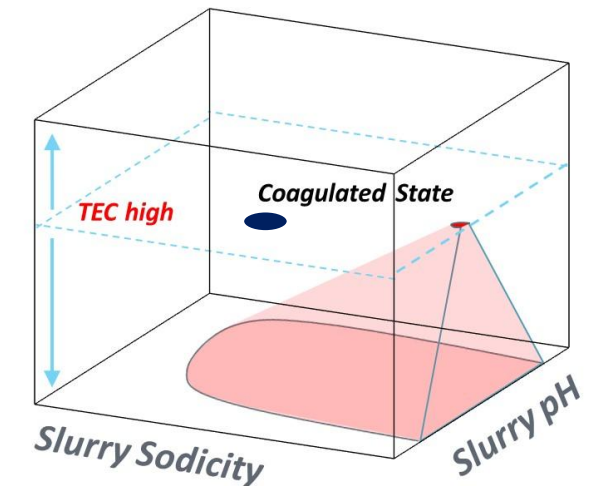
"Banana shape"
Rheological property



In waters with low Salinity (<2 mS/cm) and Sodicity >6, tailings highly dispersive



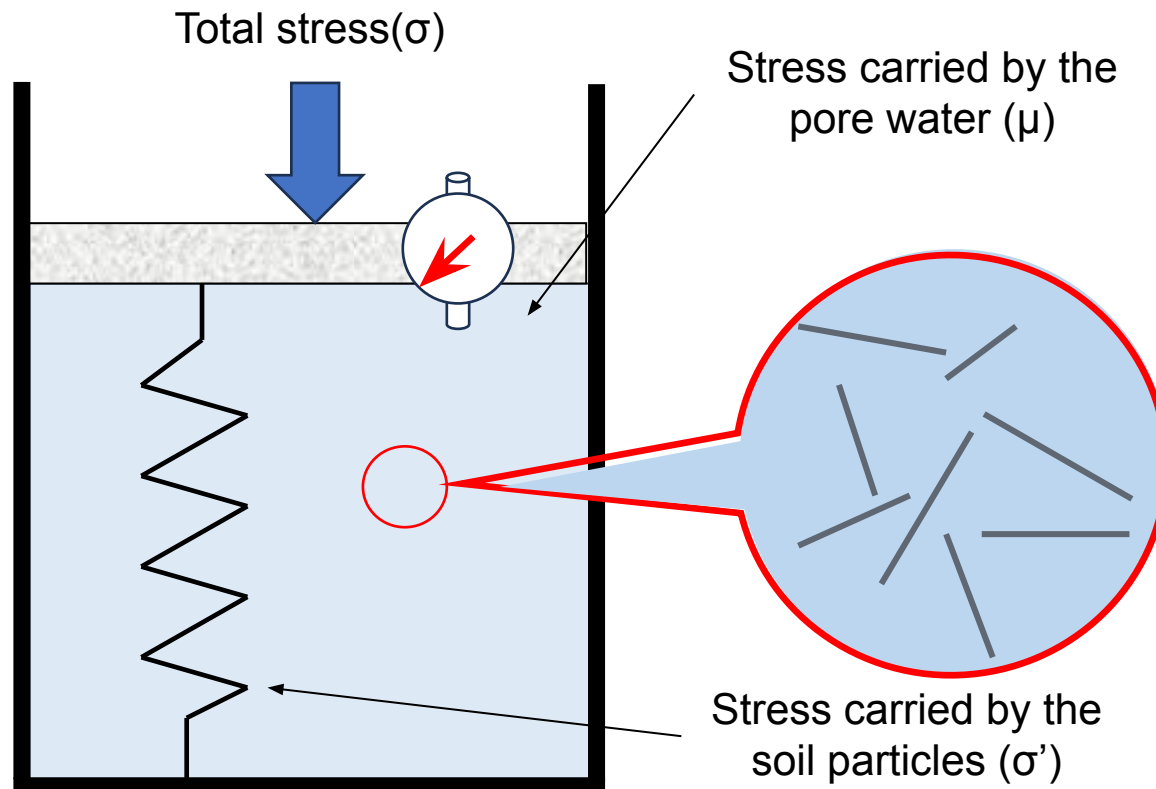
As Salinity increases only tailings with very high Sodicity remain dispersive



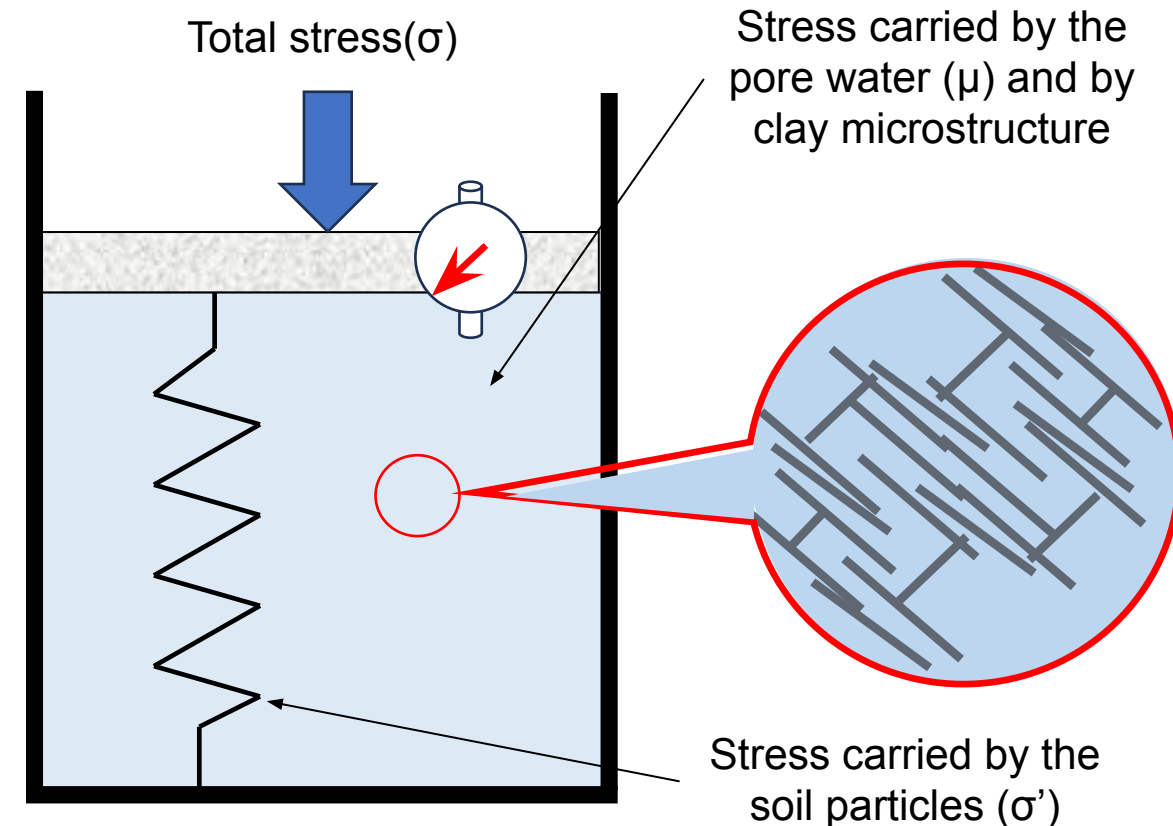
At very high Salinity (sea water) all tailings settle

Tailings Properties Overview

Dispersive Cohesive Tailings

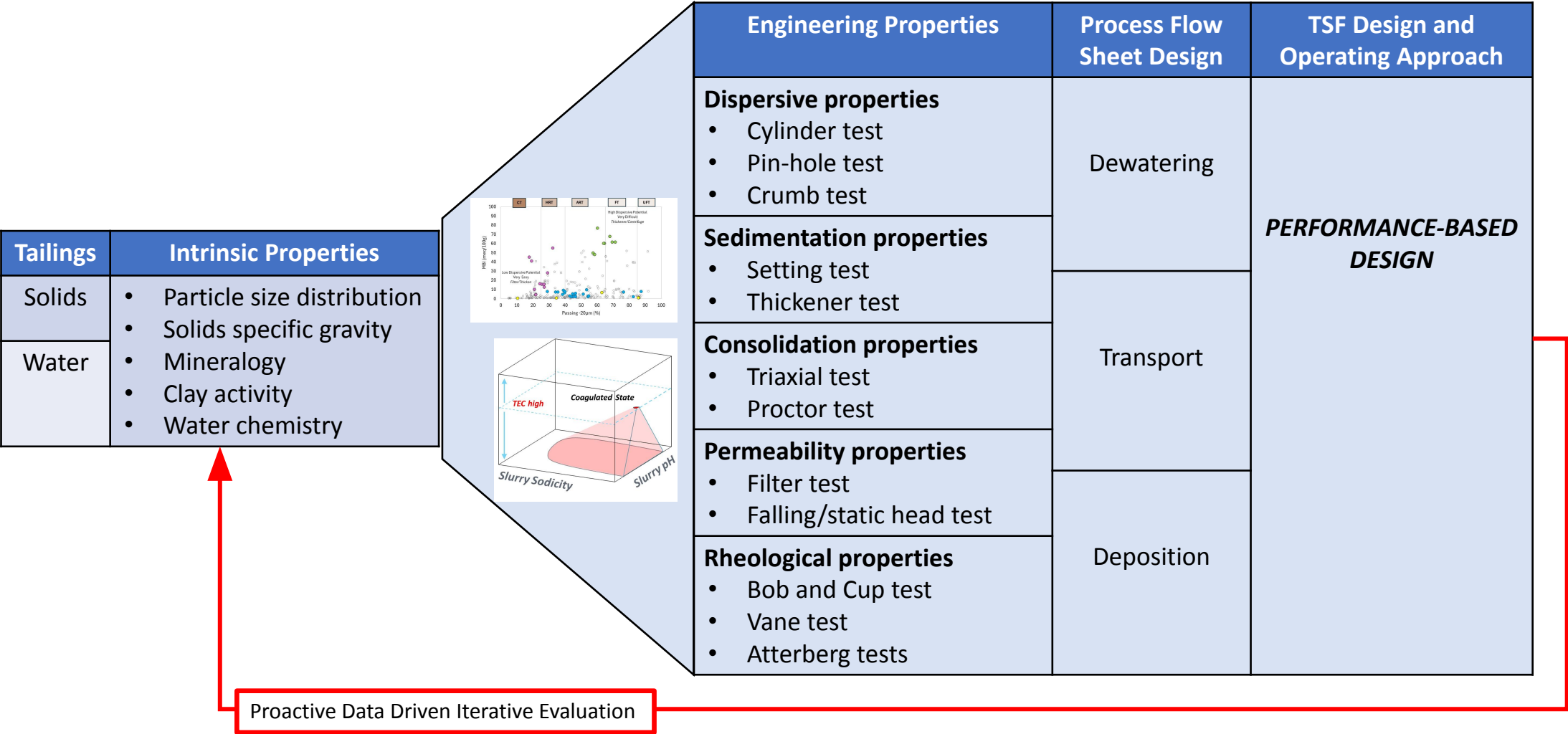


Non-dispersive Cohesive Tailings



Tailings Properties Overview

- Why classifying tailings can improve the “Observational Method” for TSF management



1. Propose a 2-tier approach for classifying tailings
 - Solids Component (psd and active clay content)
 - Water Component (dispersive properties)
2. Recognise that clay containing tailings can mimic cohesive or non-cohesive properties depending on water chemistry
3. Believe that this understanding can fill a fundamental gap in TSF design and safety